Results of a Confidential Study for the Presence of Ammonium Perfluorooctanoate in Air at a Manufacturing Facility

Barr Project No. 32/01-002-001

Prepared for Whiteman Osterman & Hanna LLP Albany, NY

April 2006

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Executive Summary

Barr Engineering Company (Barr) performed a study to determine the presence and concentration of ammonium perfluorooctanoate (APFO) in air at a confidential client facility. Study sample collection was performed on November 22, 2005. Samples were collected from the air at a facility that uses APFO containing dispersions to coat manufactured products.

Barr designed the study in cooperation with Exygen Research and the facility management, and conducted the sample collection. The samples were analyzed by Exygen Research.

Six air samples were collected and submitted for analysis, including one field blank of the sample collection media. All samples, except for the media blank sample, exhibited the presence of APFO. The American Conference of Governmental Industrial Hygienists (ACGIH) Threshold Limit Value (TLV) guideline for APFO of 0.01 mg/m³ was used in the evaluation of area air monitoring results since there were no OSHA permissible exposure levels or NIOSH recommended exposure levels to use for comparison.

The averages for air samples taken in several areas of the facility ranged from 1.2% to 62.7% of the ACGIH TLV guideline for eight-hour employee inhalation exposure to APFO. Area air samples reflected highest concentrations in manufacturing areas where APFO containing dispersions were being applied and dried. One sample taken in an area away from manufacturing activities, an employee break and lunchroom, saw the lowest air concentration of APFO in air. Representative employee exposure monitoring, conducted with APFO monitoring in locations where employees perform work activities, demonstrates that it is unlikely that the AGCIH TLV would be exceeded under similar working conditions.

Study results may be used to determine material or engineering changes to manage the presence of APFO in air and to evaluate possible changes to hygiene and work practices within the facility. They may also be used as baseline data to evaluate plant conditions at a later time under similar or alternate conditions.

Introduction

Barr Engineering Co. performed study design in cooperation with facility staff and legal advisors. An initial meeting and site survey was conducted with a Whiteman Osterman & Hanna LLP representative and facility staff on September 22, 2005. Initial determinations of the areas to be sampled and number of samples were determined. Barr provided an updated and final proposal to Scott Fein, Esq. on November 18, 2005 defining the scope or work, schedule, and cost assumptions for the study.

Sampling was performed on November 22, 2005. Specific sample locations and number of samples were finalized with plant personnel on the mobilization day previous to sampling. Tim Russell of Barr performed the sampling with coordination assistance from facility staff.

Barr collected area samples to determine APFO content in the air in various areas of the facility. The study was not designed to measure specific employee exposure to APFO. Air sample results are intended to be representative of conditions at the facility at the time that the samples were collected and may be compared to the published ACGIH TLV as a general measure of the plant air quality with respect to APFO concentration. The collection of air samples in certain areas was predicated upon activities that involve the production of aqueous fluoropolymer dispersion (AFD) coated and dried products; including mixing of AFD coatings with proprietary additives prior to application, coating of web substrates, and drying the web(s) to provide the desired properties in the finished or intermediary products.

The following table lists the general characterization of the areas sampled and the number of air samples collected.

Plant Area Characterization	Number of Air Samples
Building #4 Lunchroom	1
Building #4 Mixing Room	1
Building #4 Oven Room	1
Building #5 Oven Room	1
Building #6 Oven Room	1

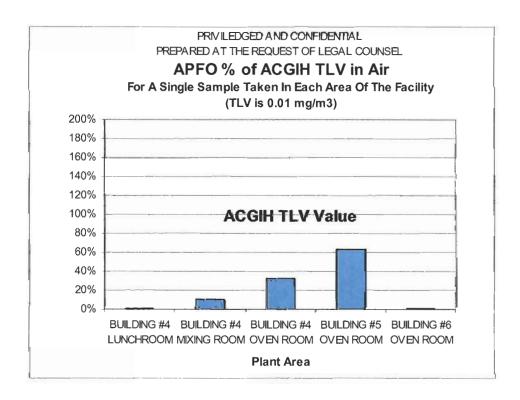
Study results may be used to determine material or engineering changes to manage the presence of APFO in air. They may also be used as baseline data to evaluate plant conditions at a later time or under various conditions.

Results and Discussion

Results of the area air monitoring samples collected for determination of APFO concentrations on November 22, 2005 are presented and discussed in this section. Specific recommendations regarding equipment, personal protective equipment (PPE), training and practices to be considered as potential improvements are also made based upon observations made during the study mobilization.

Area Air Monitoring

Table 1 presents the results of the area air monitoring. The table presents each measured APFO value, observations of facility activity and the percent relationship to the ACGIH TLV guideline. Area samples reflected highest air concentrations in manufacturing areas where APFO containing dispersions were being mixed, applied and dried. APFO air concentration in one non-manufacturing area, the Building 4 lunchroom, was the lowest measured value as illustrated in the chart below.



The plotted values are the percent relationship to the recommended TLV for the samples taken in each area. The recommended TLV is 0.01 milligrams APFO/cubic meter air (mg/m³). In addition,

the plot shows the highest value measured—62.7 percent of the recommended TLV in the Building 5 oven room. There appeared to be normal activities in all areas of the facility during the shift.

When APFO is dispersed in an aqueous mixture, its volatility is suppressed, because of the chemical's affinity to water. There is no specific temperature at which APFO begins to move into the vapor phase. Many variables in addition to temperature have been suspected to impact its volatility, such as the physical properties and other components of the mixture and the physical properties and other components of the air. When the moisture content of the air begins to increase the volatility of APFO is believed to increase. The thermal decomposition of APFO has been studied and is well documented. The rate of thermal decomposition increases with temperature. The extent of decomposition also increases with temperature. Decomposition occurs rather rapidly at 350 °C (662 °F). If held for approximately one second at 350 °C, more than 99% of the APFO would decompose. Decomposition occurs much more slowly at 196 °C (385 °F). Given that the half-life of APFO at 196 °C is 43.8 minutes [Krusic and Roe], the vapor would have to be held at that temperature for over five hours to accomplish 99% destruction.

Figure 1 provides a site map of the facility with the identification of the sample locations included in this study. Figure 2 contains photographs of the sampling equipment at each sampling location. Four of the five sampling locations were in Buildings 4 and 5, which are connected and the drying ovens in those buildings are ventilated to one common fume eliminator. Building #6 is separate and located some distance away from Buildings #4 and #5. Its drying ovens are also served by a separate ventilation system and fume eliminator.

The study results are notable in that APFO air concentrations are much higher in the Building #4 and #5 manufacturing areas (mixing and oven rooms) as compared to those in the Building #6 oven room. APFO is liberated from AFD coatings in drying and baking process steps and may be destroyed during the final sintering stage of the drying oven if the air stream is subjected to suitably high temperatures for a sufficient period of time. The Building #4 lunchroom appears to be sufficiently removed from the manufacturing areas given that it had the lowest measured concentration of APFO

#3201002001

¹ Gas-Phase NMR Technique for Studying the Thermolysis of Materials: Thermal Decomposition of Ammonium Perfluorooctanoate, Krusic, P.J. and Roe, D.C., Anal. Chem., 76:3800, 1 July 2004

² Dispersion Processor Material Balance Report, Barr Engineering, KHA Consulting, and Keller and Heckman LLP, EPA Docket on PFOA, February 2005

in air. Although observations regarding the difference in APFO ambient air concentration in Buildings #4 and #5 versus that in Building #6 are drawn from a limited number of samples, several differences in these areas may contribute to recommendations for further study or specific improvements. A table is provided below that with individual observations that may be drawn from the study results and recommendations to be considered:

	Observation	Recommendation
1	Barr speculates that the measured APFO level	Consider redesigning the mixing room
	in the mixing room air is more likely due to	ventilation system to provide local supply air
	it's location in Building #4 (immediately	ventilation and air exhaust at each mixing
	adjacent to the Building #4 oven room) than	station, rather than the area ventilation of the
	from activities in the mixing room itself.	entire workspace that is provided by the current
	APFO is not typically liberated from ambient	ceiling mounted mixing room exhaust hood.
	temperature AFD mixing processes. Though	Localized ventilation may provide the benefit
	air velocities appear low, it is likely the	of drawing less air from the oven room. Also,
	mixing room hood is drawing some air from	localizing the ventilation reduces the potential
	the adjacent oven room into the mixing room	for vented particles, mists or fumes to be drawn
	area.	across the breathing zone of employees while
		working in the mixing room.

Buildings #4 and #5 are both older than
Building #6. The Building #6 oven exhaust
ventilation system may be relatively free of
leaks that could contribute APFO to the
indoor air from the drying ovens when
compared to the Building #4 and #5 oven
exhaust ventilation systems due to differences
in building and ventilation system age and
condition. This belief is further reinforced by
the fact that the highest APFO reading in air
was measured in the Building #5 oven room,
where the ventilation ducts from Building #4
drying ovens combine with and passes
through the Building #5 oven room.

Inspect the condition of the Building #4 and #5 oven exhaust ventilation system and repair any leaks in the system duct work and ovens.

Consider studying the ventilation rates of the components of the Building #4 and #5 ventilation systems to verify that they are performing as designed and at acceptable ventilation rates at each drying oven.

Buildings #4 and #5 are served by one common ventilation and Fume Eliminator system, while Building #6 is served solely by a similar system. The exhaust from a large number of drying ovens is routed to the common Building #4 and #5 Fume Eliminator, while only three ovens are exhaust ventilated by the Building #6 Fume Eliminator. Both fume eliminators are similar in size and duct air velocities are governed by pressure differential switches that control fan speed. Thus, it is unclear why there was a lower measured APFO rate in Building 6. It may be due to the better performance of the newer coating ovens and/or greater quantity of make-up air that creates better air mixing within the room.

Inspect the condition of the Building #4 and #5 Fume Eliminator and verify its operating condition and that it is performing to design specifications. Compare the quantity of make up air added to Buildings #4 and #5 to determine if equivalent quantities of makeup air are available to operate each oven efficiently.

The measured APFO concentrations in air of this study are based on a limited number of samples. APFO air concentrations may vary during different operating scenarios, different seasonal building ventilation conditions, and at different sampling points within each area of the facility. Specific areas of potential concern include the upper areas of the drying ovens when facility staff is working to thread a new web through the oven, the ventilation ducts during periodic cleaning, and the area in and around the Fume Eliminator during maintenance and repair operations.

Consider conducting additional air monitoring to further document APFO concentrations and to provide guidance for engineering controls, changes to work processes, or respiratory protection in areas that may exceed the TLV under certain conditions.

The project was not designed to provide a thorough study of facility ventilation system(s) or in depth recommendations regarding work station improvement or modifications. However, the above observations and recommendations are intended to provide guidance for further study or actions that can be taken to assess or improve facility performance.

Facility Work, Safety and Hygiene Practices

Barr observed operations at the facility for each area undergoing air sample collection over a limited period of time at the facility and has extensive experience with similar facilities. This report is not intended to provide a comprehensive assessment of facility work or hygiene practices; however basic recommendations regarding equipment, personal protective equipment (PPE), training and practices can be made from our observations to be considered as potential workplace improvements.

General behavior in the manufacturing areas of the facility was varied with respect to safe work practices, the use of personal protective equipment, and conformance to established safety requirements. It appears that the client is providing suitable equipment and training and yet individual choice appeared to frequently guide behavior and performance. Given that, and the potential risks of variability in behavior, the client is encouraged to continue building a safety culture reinforced by all levels of management. The table below provides observations and recommendation regarding facility work, safety, and hygiene practices:

	Observation	Recommendation
1	Inconsistent use of available PPE to limit skin contact with AFD.	Fluoropolymer dispersions contain residues of fluorinated wetting agents which should not come in contact with the skin. It is necessary to wear protective gloves and other protective clothing and equipment to prevent skin contact when handling these products. Flush skin with water after contact. Enforce the use of gloves and other protective clothing or equipment during operations where the potential for skin contact exists. Require employees to practice good personal
		hygiene when leaving work area.

³ Fluoropolymers Division of the Society of Plastics Industry, "Guide to the Safe handling of Fluoropolymer Resins", 3rd Edition, June 1998, Page 35

⁴ Ibid, Page 46

2	All employees working in areas were AFD is	To prevent traces of fluoropolymer resin
	in use and the potential exists for contact	powders being carried out of the work area on
	should where uniforms during the work shift	clothing, personnel should be given the
	and change into street clothes prior to	opportunity to store their work clothing
	departing the facility. This should be	separate from their street clothing (double
	extended to include footwear as well.	locker or separate changing rooms).
		Employees should be provided with adequate
		washing facilities and encouraged to use them
		regularly. ⁵
		Wash contaminated clothing before reuse. ⁶
3	Spillage of AFD creates a slip and fall hazard.	Fluoropolymers spilled during handling should
	This was observed to occur during oven dip	be cleaned and appropriate measures should be
	pan cleaning operations.	taken to prevent the creation of a slippery
		surface. It is advisable that some form of anti-
		slip flooring or similar preventative measures
		be provided in areas where fluoropolymer
		resins are regularly handled. Slippery surfaces
		in working areas pose increased accident risks. ⁷
4	Noise levels in the Building 6 oven room	Consider performing noise monitoring to
	were perceived as high by the Barr observer.	confirm or deny that noise levels are above that
		which would require the use of hearing
		protection.

⁵ Ibid, Page 35

⁶ Ibid, Page 46

⁷ Ibid, Page 37

Although not observed during the study Staff maintaining, cleaning or repairing these mobilization, duct cleaning and Fume systems should wear appropriate PPE, Eliminator repair and maintenance activities including face splash protection, impervious likely have a potential for contact with coveralls, heavy chemical resistant gloves and concentrated residue that may contain high boot covers. Sampling of the residue may levels of APFO. moderate the level of PPE if the residue is found to have trace levels of APFO. APFO is liberated from AFD at elevated Consider measuring APFO air concentrations 6 at the elevated areas around the drying ovens temperatures. Staff working on the drying oven towers (especially near the top of the where employees work periodically. Use the tower, not at ground elevation) during sampling results to evaluate the need for operations may be exposed to APFO at levels implementing engineering controls above those measured in the study. (ventilation), changing work practices to reduce potential exposure, or providing PPE (respiratory protection) to address the potential for exposure.

Process Description

The facility consists of multiple single story buildings in the client's manufacturing campus.

Manufacturing activities that involve AFD at the campus include web coating and oven curing of the web to produce PTFE coated fabric, tapes and laminates

Aqueous fluoropolymer dispersion is mixed prior to application in one area described as the Building 4 mixing room. Buildings 4, 5, and 6 contained multiple ovens where fabric webs are coated and cured in a process that dries and bakes the webs and sinters the PTFE coating to the fabric. Facility activities are detailed below for the mixing room and oven rooms. All other areas of the building appeared to have routine operations throughout the sampling day.

Building #4 Oven Room

- Had four ovens running of which;
- Oven #7 was running 4 rolls (full capacity)
- Oven 8 was running 4 rolls (full capacity)
- Oven 10 was running 4 rolls (full capacity)
- -And Oven 11 was running one wide roll (50% capacity).

The high range of oven temperatures was in the vicinity of 725 degrees F.

All these ovens were running standard PTFE dispersions.

Building #5 Oven Room

- -Had five ovens running of which
- Ovens CB and CC were each running one roll (full capacity)
- Ovens CI and CJ were running three rolls on each (full capacity)
- And CK Oven was running one wide roll (full capacity)

The high range of oven temperatures was in the vicinity of 760 degrees F.

And ovens CB, CC, CI and CJ were running ceramic filled dispersions where as Oven CK was running standard PTFE dispersions.

Building #6 Oven Room

- Was running a standard load. No ceramic filled dispersions were being run.

The high range of oven temperatures was in the vicinity of 760 degrees F.

Mixing Room Area

- Was mixing ceramic dispersions and a drum of Mix6868 in the back mixing room and standard PTFE dispersions in the front (main) mixing room area. This is considered a normal work load for this area.

Sampling Procedures and Methods

There are no standardized methods for determination of APFO in air. An air sampling method was used for the area sampling that determined APFO concentrations by drawing air through an organic vapor sampling (OVS) tube at the rate of approximately one liter per minute. Air samples were collected during the day shift, approximately 9 AM to 4 P.M., or 420 minutes. The sample media from each sampling location was then labeled, sealed and shipped to Exygen for analysis. Concentrations were determined by dividing the reported APFO mass by the air sample volume. The result is a time-weighted concentration. The result may be compared to the published ACGIH TLV.

Sample pumps were calibrated on site before and after the shift and the arithmetic mean flow rate used to calculate the air sample volume. The OVS tubes consist of a glass housing with three separate collection media—a filter followed by two resin sorbent sections. APFO is found in particulate matter and is generally collected efficiently on a filter media. A majority of the APFO measured in the study samples was found on the filter fraction of each sample. The two resin sections following the filter in the OVS tube are analyzed separately to determine if APFO has broken-though the filter. All OVS samples had quantifiable APFO mass in the first resin section, at levels at least one order of magnitude below the measured mass on the filter. All samples showed APFO above the detection limit in the second resin section, at masses below that in the first resin section indicating efficient capture of the sample for analysis. The reported detection limit for the air analysis was 0.05 µg/fraction—samples below this value are reported as Not Detected (ND). All three air sample fractions for the media blank sample were reported as ND.

Observation notes were made about every hour to check that the pumps were operating and to document the level of manufacturing activities in the area. No problems were encountered during sample collection.

Laboratory procedures and methods are described in Exygen's report in Appendix B.

Review of Quality Assurance Data

A data validation of the Exygen Research (Exygen) data for the Ammonium Perfluorooctanoate (APFO) analyses of OVS tubes contained in the Exygen report #L0006869 was completed. The results of this validation are included in the following paragraphs.

All analysis was performed using high pressure liquid chromatography (HPLC) with a tandem mass spectrometric detector (MS/MS). In general, the areas covered by the validation process were, holding times, analytical methodology, calibration procedures, blank analysis, precision and accuracy, analytical quantitation, data package completeness, and an overall data assessment. All raw data was validated to ensure compliance in all areas.

All analytical holding times (analysis within 60 days of collection) were met, all analytical methods followed were appropriate, all calibration procedures were followed and all laboratory blanks were non-detect for all target parameters. Sample precision and accuracy were evaluated using laboratory control samples and comparing them to set criteria (70 -130% recovery for spikes and <30% relative percent difference for duplicates). All laboratory control samples met all acceptance criteria. All analytical identification and quantitation procedures were evaluated and no quantitation errors were detected. An isotopically labeled perflurocatanoic acid surrogate was spiked on each OVS tube prior to sample extraction and analysis. The percent recoveries of this compound from each sample were compared to established criteria (70-130%). One sample (Building 5 Oven T1R4) had a percent recovery (169%) above the set criteria. Based on the review of the data, it was determined that this criteria exceedance was due to the level of APFO present in the spiked sample and does not reflect a larger quality assurance issue. No qualification of the data was performed based on the data review process.

Overall the laboratory data met the project requirements. The final data met the project data quality objectives and should be considered acceptable as reported.

Table 1 Result for APFO Concentration in Area Air Samples

TABLE 1

Confidential Study with Exygen Research Result for APFO Concentration in Area Air Samples

Percent of TLV^1	Sampler located above lunchroom refrigerator @ 0853 1.20% No employees present @ 1015, 1118, and 1515 Four employees present at 1212, five employees @1418, and 2 employees @ 1553	Sampler located on workstation serving mixing room @ 0857 10.94% No employees present with two mixing stations operating @ 1022 No employees present with three mixing stations operating @ 1333, 1422, and 1517 One employee present with two mixing stations operating @ 1214 One employee present with three mixing stations operating @ 1120 and 1558	Sampler located at end of Oven 8 ~ 3 feet above dip pan @ 0902 32.95% Ovens 7, 8, 10, 11 operating during the shift. Between 3-6 employees in area during the shift White dust of unknown origin noted near sampling pump.	Sampler located on workstation between Oven CF and CK @ 0910 62.70% Ovens CB, CC, CI, CJ and CK operating during the shift Between 3-6 employees in area during the shift	Sampler located on Oven B workstation @ 0924 1.46% Ovens A, B, and C operating during the shift Two employees in area during the shift
APFO Concentration mg/m³	1.2E-04	1.1E-03	3.3E-03	6.3E-03	1.5E-04
Elapsed Time min	420	421	420	421	420
Average Calibration I/min	1.0037	0.9974	0.9992	0.9980	1.0024
APFO Mass/ sample	0.05043	0.45954	1.38271	2.63431	0.06128
Sample Location	BUILDING 4 LUNCHROOM	BUILDING 4 MIXING ROOM	BUILDING 4 OVEN ROOM	BUILDING 5 OVEN ROOM	BUILDING 6 OVEN ROOM

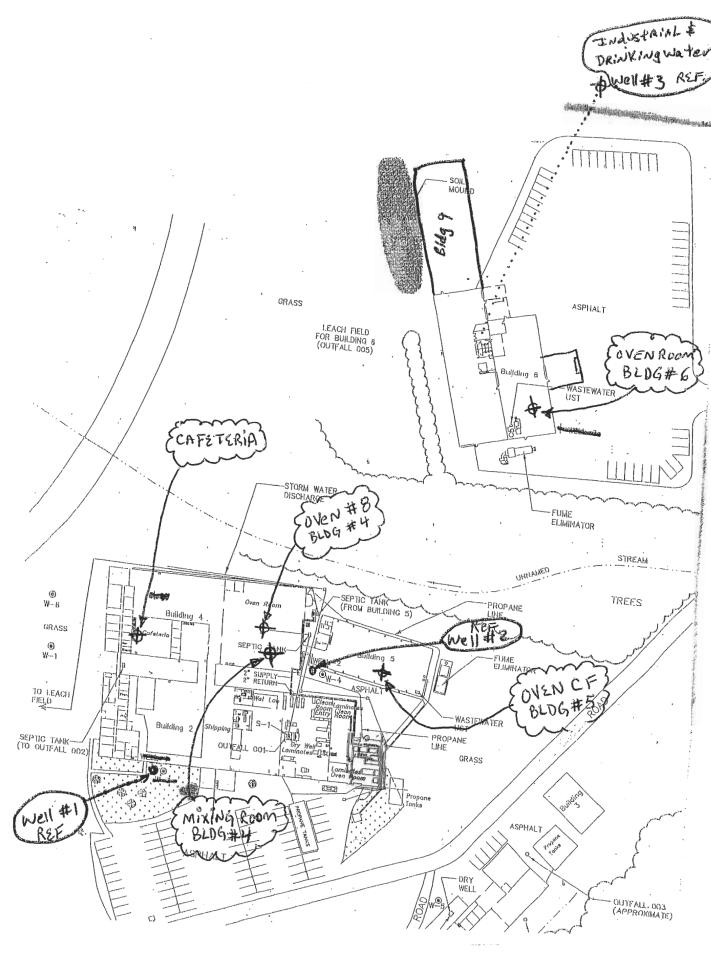
¹Threshold Limit Value, 0.01 mg/m³

2

Media Blank

 2 Observations were spaced roughly hourly for entire shift ND = Not Detected, response between 0 and 0.05 $\mu g/sample$ fraction

Figure 1 Site Map and Sample Locations



Ambient Air Sampling Locations November 72, 2005 04645

Figure 2 Photographs of Air Sampling Locations











Appendices

Appendix A

Report Calculations and Nomenclature

Confidential Study with Exygen Research **Calculations for APFO Concentration in Area Samples**

			APFO						
	Lo	APFO Mass/	Mass/	Pretest	Posttest	Average	Elapsed	APFO	Percent
Sample Location	Fraction	fraction	sample	Calibration	Calibration	Calibration	Time	Concentration	of TLV ¹
	<u> </u>	ng/fraction	μg	l/min	I/min	I/min	min	mg/m ³	
BUILDING 4 LUNCHROOM									
	Α	46.4	0.05043	1.008	0.9993	1.0037	420	1.2E-04	1.20%
	В	2.72							
	С	1.31							
BUILDING 4 MIXING ROOM									
	Α	452	0.45954	1.011	0.9837	0.9974	421	1.1E-03	10.94%
	В	6.01							
	С	1.53							
BUILDING 4 OVEN ROOM									
	Α	1370	1.38271	1.008	0.9904	0.9992	420	3.3E-03	32.95%
	В	10.5							
	С	2.21							
BUILDING 5 OVEN ROOM									
	Α	2600	2.63431	1.009	0.9869	0.9980	421	6.3E-03	62.70%
	В	30.4							
	С	3.91							
BUILDING 6 OVEN ROOM									
	Α	49.3	0.06128	1.008	0.9967	1.0024	420	1.5E-04	1.46%
	В	9.42							
	С	2.56							
Media Blank	Α	ND	ND						
	В	ND							
	С	ND							

 1 Threshold Limit Value, 0.01 mg/m 3 ND = Not Detected, response between 0 and 0.05 $\mu g/fraction$

Appendix B

Laboratory Reports and Sample Chain of Custody



Analytical Report

Determination of Ammonium Perfluorooctanoate (APFO) in OVS Airtube Samples

Exygen Report No. L0006869

Testing Laboratory

Exygen Research 3058 Research Drive State College, PA 16801

Requester

Tim Russell
BARR Engineering Co.
4700 West 77th Street
Minneapolis, MN 55435-4803
Phone: 952-832-2600



1 Introduction

Results are reported for the analysis of ammonium perfluorooctanoate (APFO) in OVS airtube samples received at Exygen from BARR Engineering. The Exygen project number assigned to the samples is L0006869. Table I lists the analyte quantitated for the samples.

Table I. Target Analytes

<u>Parameter</u> <u>Acronym</u>

Ammonium APFO

Perfluorooctanoate

2 Sample Receipt

Six OVS airtube samples were received at Exygen. The OVS airtube samples were split into three fractions each. A total of twenty-one samples were analyzed, including laboratory controls. A copy of all sample log-in information is presented in Attachment C.

The samples were received on November 29, 2005. The samples were delivered on ice by FEDEX. Exact shipping details can by found in Attachment C. The samples were stored refrigerated from time of receipt until analysis.

3 Holding Times

The maximum holding time for completion of analysis is sixty days from sample collection. All holding times were met for this data set.

4 Methods - Analytical and Preparatory

4.1 OVS Airtube Sample Extraction

All seven OVS airtubes were fortified with 1 ng of 13 C PFOA by spiking 10 μ L of a 100 ng/mL solution directly onto the glass fiber frit. The airtubes were allowed to air-dry at room temperature for several minutes before being disassembled for the extraction. For extraction, the airtubes were disassembled into three fractions, with each fraction being placed into individual 20mL glass scintillation vials. Fraction A consisted of the plastic retaining ring and glass fiber frits. Fraction B consisted of the first section of XAD resin beads and the first PUF filter. Fraction C consisted of the last section of XAD resin beads and PUF filter. Next, 1.0 mL of HPLC grade methanol was added to each vial. Each sample was mixed for ~30 seconds. The

process was repeated two more times at ~10 minute intervals. The extract was then withdrawn from the vials using disposable glass pipettes and transferred to 2mL HPLC vials. The extracts were analyzed using electrospray LC/MS/MS.

4.2 Sample Analysis by LC/MS/MS

In High Pressure Liquid Chromatography (HPLC), an aliquot of extract is injected and passed through a liquid-phase chromatographic column. Based on the affinity of the analyte for the stationary phase in the column relative to the liquid mobile phase, the analyte is retained for a characteristic amount of time. Following HPLC separation, mass spectrometry provides a rapid and accurate means for analyzing a wide range of organic compounds. Molecules are ionized, fragmented, and detected. The ions characteristic of the compounds are observed and quantitated against extracted standards.

An HP1100 system interfaced to a Micromass Quattro system was used to analyze the sample extracts for quantitation. A gradient elution through a Jones Chromatography Genesis C-8 $50 \times 2.1 \text{ mm} \times 4 \mu \text{m}$ column was used for separation.

The following gradient was performed:

Mobile Phase (A): Mobile Phase (B):	2mM Ammonium Acetate in Water Methanol			
<u>Time</u>	<u>%A</u>	<u>%B</u>		
0.0	60	40		
1.0	10	90		
7.0	10	90		
7.5	0	100		
9.0	0	100		
9.5	60	40		
13.5	60	40		
14.0	60	40		
15.0	60	40		

The following parameters were used for operation of the mass spectrometer:

Parameter	Setting
Ionization Mode	Electrospray
Polarity	Negative
Transitions Monitored	413->369 (PFOA); 415->370 (13C PFOA)
Gas Temperature	350°C
Drying Gas (N2)	7.0 L/min

5 Analysis

5.1 Calibration

An 8-point calibration curve was analyzed throughout the analytical sequence for PFOA. The calibration points were prepared at 0.05, 0.1, 0.2, 0.5, 1.0, 5.0, 10, and 50 ng/mL (ppb) for LC/MS/MS analysis. The instrument response versus the concentration was plotted for each point. Using linear regression with 1/x weighting, the slope, y-intercept and coefficient of determination (r^2) were determined. A calibration curve is acceptable if $r^2 \ge 0.985$.

For the results reported here, calibration criteria were met. The calibration curves are included in the raw data in Attachment C.

5.2 Surrogates

For OVS airtube samples, ¹³C PFOA was added to the top of the tube in the laboratory before the extraction. ¹³C PFOA is used as a surrogate for OVS airtube samples.

¹³C PFOA recoveries can be found in Attachment B.

5.3 Laboratory Control Spikes

The laboratory control spike in the analytical set was prepared by adding a known concentration of the analyte to a laboratory control airtube. Laboratory control spikes are used to assess method accuracy. The laboratory control spike must show recoveries between 70-130% or the data is rejected. For the results reported here, the laboratory control spikes were within the acceptable range.

5.4 Matrix Spikes

There were no matrix spike samples in the set of OVS airtubes.

6 Data Summary

Please see Attachment A for a detailed listing of the analytical results. Results are reported in ng/fraction for the analyte for OVS airtube samples on an as-received basis.

7 Data/Sample Retention

Samples are disposed of one month after the report is issued unless otherwise specified. All electronic data is archived on retrievable media and hard copy reports are stored in data folders maintained by Exygen. Hardcopy data is stored for a minimum of five years. The client will be notified 30 days prior to the disposal of hardcopy data.

8 Attachments

- 8.1 Attachment A: Chain of Custody
- 8.2 Attachment B: Analytical Results
- 8.3 Attachment C: Raw Analytical Data

9 Signatures

Chrotine E Edwards	12/22/05	
Christine Edwards, Laboratory Technician	Date	
Olm Plated	15/25/05	
John M. Flaherty, Vice President	Date	





Summary of APFO in OVS Tubes

Analyte Found (ng/fraction)

Sample ID	Analyte Found (ng/fraction) APFO
Sample Media Blank T1R0 A	ND
Sample Media Blank T1R0 B	ND
Sample Media Blank T1R0 C	ND
Duilding 4 office T4D4 A	46.4
Building 4 office T1R1 A	46.4
Building 4 office T1R1 B	2.72
Building 4 office T1R1 C	1.31
Building 4 mixing T1R2 A	452
Building 4 mixing T1R2 B	6.01
Building 4 mixing T1R2 C	1.53
	1.00
Building 4 oven T1R3 A	1370
Building 4 oven T1R3 B	10.5
Building 4 oven T1R3 C	2.21
Building 5 oven T1R4 A	2600
Building 5 oven T1R4 B	30.4
Building 5 oven T1R4 C	3.91
Building 6 oven T1R5 A	49.3
Building 6 oven T1R5 B	9.42
Building 6 oven T1R5 C	2.56

ND = Not Detected. Response between 0 and 0.05 ng/fraction.



Recovery Summary for PFOA in OVS Tubes

Sample ID	Analyte Added (ng/mL) PFOA	Analyte Found (ng/mL) PFOA	Recovery
Control 1 A Spk	2.0	1.81	91



Recovery Summary for ¹³C-PFOA in OVS Tubes

Sample ID	Analyte Added (ng/mL) 13C-PFOA	Analyte Found (ng/mL) 13C-PFOA	Recovery (%)
Sample Media Blank T1R0 A	1.0	0.948	95
Control 1 A	1.0	0.877	88
Building 4 office T1R1 A	1.0	0.896	90
Building 4 mixing T1R2 A	1.0	0.923	92
Building 4 oven T1R3 A	1.0	1.30	130
Building 5 oven T1R4 A	1.0	1.69	169
Building 6 oven T1R5 A	1.0	0.839	84



3058 Research Drive State College, PA 16801

Phone: 814-272-1039 Fax: 814-231-1580

True

True

True

True

Conform COC Sample:

Conform COC:

Conform Sample:

Conform Request:

Login

Login Group: L0006869

Login #:

6980

Project:

P0001903

Company Name: Barr Engineering

Submitted By: Login Type:

Tim Russell Immediate Receipt of Samples

Started:

True

11/30/2005

Login.rpt

Date Start:

11/30/2005 12/10/2005

Due Date: Received Date: Received By:

11/29/2005 Ammerman, Mark

Spread Sample:

Label:

Exygen SD/PI:

Risha, Karen

Project Title/Type: Analysis of APFO in OVS Tubes by LCMSMS / ROUTINE

Login Notes:

Packages / Containers

<u>Package</u>	Carton	Date / Co	ndition	Shipper / ID	Temp. Control/Temp.	Direction / Handled By
PK0007957		Date: 11/2 Contents U	9/05 10:18 Incompromised	FEDEX 4532 9210 4227	Wet Ice 4.8	RECEIVED Ammerman, Mark
Container #	Gross Weight) Hg	Container Type	<u>Preservative</u>	Mfg. Lot	Mfg. ID
C0123806	6.70 g		Air tube	NONE		
C0123807	6.70 g		Air tube	NONE		
C0123808	6.70 g		Air tube	NONE		
C0123809	6.50 g		Air tube	NONE		
C0123810	6.50 g		Air tube	NONE		
C0123811	6.50 g		Air tube	NONE		
C 0123812	6.50 g		Air tube	NONE		
C0123813	6.50 g		Air tube	NONE		
C0123814	6.50 g		Air tube	NONE		
C0123815	6.50 g		Air tube	NONE		
C0123816	6.50 g		Air tube	NONE		
C0123817	6.50 g		Air tube	NONE		
C0123818	6.60 g		Air tube	NONE		
C0123819	6.60 g		Air tube	NONE		
C0123820	6.60 g		Air tube	NONE		
C0123821	6.60 g		Air tube	NONE		
C0123822	6.60 g		Air tube	NONE		
C0123823	6.60 g		Air tube	NONE		
C0123824	6.50 g		Air tube	NONE		
C0123825	6.50 g		Air tube	NONE		
C0123826	6.50 g		Air tube	NONE		

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Instance:

Login

				Samples		
<u>Sample ID</u> L0006869-0001	Container C0123806	<u>Matrix</u> GAS	<u>Fraction</u> OVS Tubes	Sample Sample Media Blank T1R0 A	<u>Date Sampled</u> 11/22/2005	<u>Date Due</u> 12/10/2005
L0006869-0002	C0123807	GAS	OVS Tubes	Sample Media Blank T1R0 B	11/22/2005	12/10/2005
L0006869-0003	C0123808	GAS	OVS Tubes	Sample Media Blank T1R0 C	11/22/2005	12/10/2005
L0006869-0004	· C0123809	GAS	OVS Tubes	Building 4 office T1R1 A	11/22/2005	12/10/2005
L0006869-0005	C0123810	GAS	OVS Tubes	Building 4 office T1R1 B	11/22/2005	12/10/2005
L0006869-0006	C0123811	GAS	OVS Tubes	Building 4 office T1R1 C	11/22/2005	12/10/2005
L0006869-0007	C0123812	GAS	OVS Tubes	Building 4 mixing T1R2 A	11/22/2005	12/10/2005
L0006869-0008	C0123813	GAS	OVS Tubes	Building 4 mixing T1R2 B	11/22/2005	12/10/2005
L0006869-0009	C0123814	GAS	OVS Tubes	Building 4 mixing T1R2 C	11/22/2005	12/10/2005
L0006869-0010	C0123815	GAS	OVS Tubes	Building 4 oven T1R3 A	11/22/2005	12/10/2005
L0006869-0011	C0123816	GAS	OVS Tubes	Building 4 oven T1R3 B	11/22/2005	12/10/2005
L0006869-0012	C0123817	GAS	OVS Tubes	Building 4 oven T1R3 C	11/22/2005	12/10/2005
L0006869-0013	C0123818	GAS	OVS Tubes	Building 5 oven T1R4 A	11/22/2005	12/10/2005
L0006869-0014	C0123819	GAS	OVS Tubes	Building 5 oven T1R4 B	11/22/2005	12/10/2005
L0006869-0015	C0123820	GAS	OVS Tubes	Building 5 oven T1R4 C	11/22/2005	12/10/2005
L0006869-0016	C0123821	GAS	OVS Tubes	Building 6 oven T1R5 A	11/22/2005	12/10/2005
L0006869-0017	C0123822	GAS	OVS Tubes	Building 6 oven T1R5 B	11/22/2005	12/10/2005
L0006869-0018	C0123823	GAS	OVS Tubes	Building 6 oven T1R5 C	11/22/2005	12/10/2005
L0006869-0019	C0123824	GAS	OVS Tubes	Control 1 A		12/10/2005
L0006869-0020	C0123825	GAS	OVS Tubes	Control 1 B		12/10/2005
L0006869-0021	C0123826	GAS	OVS Tubes	Control 1 C		12/10/2005

Instance: TAC FPA162662

Login

Login Reviewed By: 13m STT CWA Date/Time: 12/5/05 1540

Report Version: Aug 24 2005 3:47PM Page 3 of 3

Instance: TAC F6311357663



CHAIN OF CUSTODY/ANALYSIS REQUEST FORM

Exygen Research Sample Receiving • 3048 Research Drive • State College, PA 16801, USA T: 814.231.8032 • F: 814.231.1580 • exygenresearch.com

Page / of /

ANALYSES REQUESTED	51	n/5)	w/>	7	C	2101		*	×	×	×	· · ·	×
	e & E-mail Address):	yosterman & Hama CLL	From Jay Bronowing	andling of your samples.			Comments	Analyze three Fractions	of 005 TUBE (F) HEr	and two soveent	sections) segantely,	1	
	Project Manager (Name & E-mail Address):	Project Name:	P.O. #: Quotation #: [[mail] Fran	· —		posite ber of sainers	Grat Com	x 1 0VS	200 x	\ \ \ \ \ \ \	5VO x .	× 0√5	× × ×
PROJECT INFORMATION	Client (name & address): BARK ENGINEERING CO. 7390 OFWE LAND	£8£2-258-25b	125-228-256 120521	ise fill out this form <i>completely</i> to ensure		SAMPLE ANALYSIS	Client Sample Identification Date Time	SAMPLE MOIS BRAK TIRO 11-22-05 1553	Bulding 4 OFFICE TIRI 7 1553	Contiding of mixing TIRZ 1557	Building 4 oven TIR3 (1602	Building 5 oven TIRY 1610	Building 6 oven TI 125 V 1624
PROJECT	Client (n	Phone:	Fax: Sampler:	P.		SAM	ExyLIMS#						

Time
Date
Relinguished by

LAB USE ONLY

Relinquished by,	Date	Time
11 / Tim 12455EI)	50-82-11	0800

	Received by	Date	Time
		11/02/14	3/0/
_			
_			
$\overline{}$			
	LAB USE ONLY		

Add case narrative

Add quality control summary

Add calibration summary

Add raw data

Other

PROJECT REQUIREMENTS

-aboratory Report Options:

X Sample results only

10 WOLKINDONS

Results Deadline:

Cooler Temp. (°C)

Cooler ID #

9EV. 6. ^ .

OTHER INFORMATION

TAC EPA 04664



3058 Research Drive Phone: 814-272-1039 State College, PA 16801 Fax: 814-231-1580

TEMPORARY SAMPLE STORAGE FORM

To be completed during ExyLIMS Login
Project #: <i>P</i> 1903
Login #:
Initials / Date: Mr4 11/3d-r
One form to be completed for each package
Date / Time Received: / ///29 / /0/%
Received By: Mak Amore
Shipper: Folk
Shipper Package ID: 4534-9210 4227
Temperature (deg C) / Thermometer ID: 4.8 Davi 77 P. Mill Illaylor
Temperature Control Method:
Temporary Storage Location:
Condition of sample(s): Good – Package and contents uncompromisedFair – Package damaged / contents uncompromised
Poor – Package and contents compromised Notes:

LIBRARY ID: V0001402-3

LABORATORY FORM

ORIGIN ID: BBBA (612) 832-2600 BARR ENGINEERING COMPANY BARR ENGINEERING COMPANY 4700 WEST 77TH STREET SUITE 200 EDINA, MN 554354803 UNITED STATES US Ship Date: 28NOV05 Actual Wgt: 3.7 LB System#: 0081929/CAFE2285 Account: S 055429979

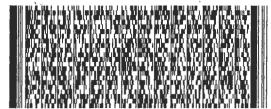
TO KAREN RISHA--SAMPLE RECEIVING EXYGEN RESEARCH 3048 RESEARCH DRIVE

(814) 231-8032 FedEx

STATE COLLEGE, PA 16801

Express

REF: 3201002001002THRLTS



Delivery Address Barcode

BILL SENDER

PRIORITY OVERNIGHT

TRK# **4532 9210 4227**

Form 7 0201 TUE
Deliver By
29NOV05
IT AA

16801 -PA-US

Part # 156148 42: NGT P 23

NV SCEA

6ho 14 4.8

RAW DATA REPORT

Sponsor Study No: Exygen Study No: Analyte: Ions Monitored:

NA L6869 APFO 413 -> 369 Limit of Quantitation: Injection Volume: Matrix: Sample Volume Extracted:

Final Volume:

0.1 ng/mL 15 µL **OVS Tubes** NA 1.0 mL

Set No: Analyst: Instrument Type: Extraction Date: Analyzed on:

120205B Chrissy Edwards LC/MS/MS Unit # 6 12/02/05 12/17/05

				Std.			PFOA	PFOA	APFO	Amount PFOA	
Exygen	Sponsor	Sample	Run	Conc.	Dilution	Peak	Found	Found	Found	Added	Re
ID	ID	Code	No.	(ng/mL)	Factor	Area	(ng/mL)	(ng/fraction)	(ng/fraction)	(ng/mL)	K
C120205-8	•	CS	120205B-101	0.05	-	6850	("5"1127	-	-	-	_
C120205-7	-	CS	120205B-102	0.1	-	9763	-	-	-	-	
C120205-6	-	CS	120205B-103	0.2		19864			-		
C120205-5	-	CS	120205B-104	0.5	-	56506	-	_			
C120205-4	-	CS	120205B-105	1.0		84296	-				
C120205-3	_	CS	120205B-106	5.0	_	324180					
C120205-2		CS	120205B-107	10.0	_	612627	-	-			
C120205-1		CS	120205B-108	50.0	_	2604609	-		_	_	
Methanol Wash		c	120205B-109			2631			_	_	
L6869-1 Blank	Sample Media Blank T1R0 A	c	120205B-110	_	ı	2579	ND	ND	ND	-	
L6869-2 Blank	Sample Media Blank T1R0 B	Č	120205B-111		i	2836	ND	ND	ND	_	
L6869-3 Blank	Sample Media Blank T1R0 C	Č	120205B-112	_	1	1850	ND	ND	ND		
L6869-19 Blank Spk	Control 1 A	LCS	120205B-I13	_	1	103451	1.81	1.81	1.88	2.0	
L6869-20 Blank Spk	Control 1 B	LCS	120205B-114		1	6265	ND	ND	ND	_	
L6869-21 Blank Spk	Control I C	LCS	120205B-115	_	1	2492	ND	ND	ND		
C120205-8		CS	120205B-116	0.05		6456	-				
C120205-7	_	CS	120205B-117	0.1		10733	_	_			
L6869-4	Building 4 office TIR1 A	S	120205B-118	•	1	2384639	44.6	44.6	46.4		
L6869-5	Building 4 office T1R1 B	S	120205B-110		1	146376	2.61	2.61	2.72		
L6869-6	Building 4 office T1R1 C	S	120205B-119	-	1	74092	1.26	1.26	1.31	-	
L6869-7	Building 4 mixing T1R2 A	S	120205B-120		10	2322318	434	434	452	-	
L6869-7	Building 4 mixing T1R2 A	S	120205B-121	-	1	16821050	*	-	432	•	
L6869-8	Building 4 mixing T1R2 B	S	120205B-122		1	315051	5.77	5.77	6.01	-	
L6869-9	Building 4 mixing T1R2 C	S	120205B-123		1	85547	1.47	1.47	1.53	•	
C120205-6	Building 4 mixing TTR2 C	CS	120205B-125	0.2	-	18159	-	-	1,33	-	
C120205-5		CS	120205B-126	0.5	-	54222	-	-		-	
L6869-10	Building 4 oven T1R3 A	S	120205B-120	-	100	710837	1320	1320	1374	•	
L6869-10	Building 4 oven T1R3 A	S	120205B-127		1	34356384	1320 •	1320	1374	•	
L6869-11	Building 4 oven T1R3 B	S	120205B-129	-	1	544090	10.1	10.1	10.5	•	
L6869-12	•	S	120205B-129		I I	119943	2.12	2.12		-	
L6869-12 L6869-13	Building 4 oven T1R3 C	S	120205B-130		100	1339684	2500	2500	2.21 2603	-	
L6869-13	Building 5 oven T1R4 A	S	120205B-131		1	49394532	2300 *	2300	2603	•	
L6869-14	Building 5 oven T1R4 A Building 5 oven T1R4 B	S	120205B-132		1	1563750	29.2	29.2	30.4	-	
	_				-					•	
L6869-15	Building 5 oven T1R4 C	S	120205B-134		1	207540	3.76	3.76	3.91	-	
Methanol Wash	•	С	120205B-135	-	-	2549	•	-	-	-	
C120205-4	-	CS	120205B-136	1.0		75278	-	-	-	•	
C120205-3	D DE COMMEN	CS	120205B-137	5.0	-	319383		-	-	•	
L6869-16	Building 6 oven T1R5 A	S	120205B-138	-	100	32292	47.4	47.4	49.3	-	
L6869-16	Building 6 oven T1R5 A	S	120205B-139	-	1	2680607		-	-	-	
L6869-17	Building 6 oven T1R5 B	S	120205B-140	-	1	489932	9.05	9.05	9.42	-	
L6869-18	Building 6 oven T1R5 C	S	120205B-141		1	138052	2.46	2.46	2.56	-	
C120205-2 C120205-1	-	CS	120205B-142	10.0	-	555354	•	-	-	-	
		CS	120205B-143	50.0	-	2488549		-	-	-	

Recovery (%) =

S = Sample

[PFOA found (ng/mL)-PFOA found in control (ng/mL)] x 100 amount PFOA added (ng/mL)

PFOA Found (ng/fraction) =(PFOA Found (ng/mL) x 1 mL)

APFO Found (ng/fraction) = PFOA found (ng/fraction) x (MW APFO (431) / MW PFOA (414))

CS = Calibration standard C = Control sample

LF = Lab fortified sample

LCS = Laboratory Control Spike

ND = Not detected = Response between 0 and 0.05 ng/mL

NQ = Not quantifiable = Response between 0.05 and 0.1 ng/mL

*Sample was analyzed at several dilution levels in this set. The appropriate result is reported.

Spreadsheet prepared by: CFE 12 22 05

Slope = 53365.6 Coef. Of Det. = 0.995131

RAW DATA REPORT

Sponsor Study No: Exygen Study No: Analyte:

NA L6869 13C-PFOA 415 -> 370

Limit of Quantitation: Injection Volume:

0.1 ng/mL 15 μL OVS Tubes Set No: Analyst: Instrument Type:

120205B Chrissy Edwards LC/MS/MS Unit # 6

Ions Monitored: Site:

NA

Sample Volume Extracted: Final Volume:

Matrix:

NA 1.0 mL Extraction Date: Analyzed on:

12/02/05 12/17/05

								Amount	
_				Std.	D'1 .'	n 1	13C-PFOA	13C-PFOA	
Exygen	Sponsor	Sample	Run	Conc.	Dilution	Peak	Found	Added	Recovery
ID	ID	Code	No.	(ng/mL)	Factor	Area	(ng/mL)	(ng/mL)	(%)
C120205-8	-	CS	120205B-101	0.05	-	4538	-	-	-
C120205-7	-	CS	120205B-102	0.1	-	7649	-	-	-
C120205-6	-	CS	120205B-103	0.2	-	16228	-	-	-
C120205-5	-	CS	120205B-104	0.5	-	48920	-	-	-
C120205-4	-	CS	120205B-105	1.0	-	69840	-	-	-
C120205-3	-	CS	120205B-106	5.0	-	293826	-	-	-
C120205-2	-	CS	120205B-107	10.0	-	519815	-	-	-
C120205-1	-	CS	120205B-108	50.0	-	2362054	-	-	-
Methanol Wash	-	С	120205B-109	-	-	0	-	-	-
L6869-1 Blank	Sample Media Blank T1R0 A	С	120205B-110	-	1	50054	0.948	1.0	95
L6869-2 Blank	Sample Media Blank T1R0 B	С	120205B-111	-	1	1018	ND	-	-
L6869-3 Blank	Sample Media Blank T1R0 C	C	120205B-112	-	1	0	ND	-	-
L6869-19 Blank Spk	Control 1 A	LCS	120205B-113	-	1	46675	0.877	1.0	88
L6869-20 Blank Spk	Control I B	LCS	120205B-114	-	1	265	ND	-	-
L6869-21 Blank Spk	Control 1 C	LCS	120205B-115	-	1	0	ND	-	-
C120205-8	-	CS	120205B-116	0.05	-	4573		-	-
C120205-7	-	CS	120205B-117	0.1	-	7800	-	-	-
L6869-4	Building 4 office T1R1 A	S	120205B-118	-	1	47574	0.896	1.0	90
L6869-5	Building 4 office T1R1 B	S	120205B-119	, -	1	663	ND	-	-
L6869-6	Building 4 office T1R1 C	S	120205B-120	-	1	272	ND	-	-
L6869-7	Building 4 mixing T1R2 A	S	120205B-121	-	10	6750	*	1.0	-
L6869-7	Building 4 mixing T1R2 A	S	120205B-122	-	1	48873	0.923	1.0	92
L6869-8	Building 4 mixing T1R2 B	S	120205B-123	-	1	1046	ND	-	-
L6869-9	Building 4 mixing T1R2 C	S	120205B-124	-	1	209	ND	-	-
C120205-6	-	CS	120205B-125	0.2	-	14355		_	-
C120205-5	_	CS	120205B-126	0.5	_	46496		-	_
L6869-10	Building 4 oven T1R3 A	S	120205B-127	-	100	1246	*	1.0	_
L6869-10	Building 4 oven T1R3 A	S	120205B-127	_	1	66853	1.30	1.0	130
L6869-11	Building 4 oven T1R3 B	S	120205B-120	_	i	1047	ND	-	
L6869-12	Building 4 oven T1R3 C	S	120205B-129	-	1	324	ND	-	
L6869-13	Building 5 oven T1R4 A	S	120205B-130		100	1771	*	1.0	_
L6869-13	Building 5 oven T1R4 A	S	120205B-131	-	1	85820	1.69	1.0	169
L6869-14	Building 5 oven T1R4 B	S	120205B-132		1	1923	ND	-	102
L6869-15	_	S	120205B-133		1	457	ND	-	-
	Building 5 oven T1R4 C	C		-	-	95	ND	-	•
Methanol Wash	-	CS	120205B-135	1.0	-	64264		•	-
C120205-4	-		120205B-136				-	-	-
C120205-3		CS	120205B-137	5.0	-	284574	*	-	-
L6869-16	Building 6 oven T1R5 A	S	120205B-138	-	100	525		1.0	0.4
L6869-16	Building 6 oven T1R5 A	S	120205B-139	-	1	44847	0.839	1.0	84
L6869-17	Building 6 oven T1R5 B	S	120205B-140	-	1	567	ND	-	-
L6869-18	Building 6 oven T1R5 C	S	120205B-141	-	1	173	ND	-	•
C120205-2	-	CS	120205B-142	10.0	•	471436	-	-	•
C120205-1	-	CS	120205B-143	50.0	-	2265808	-	-	-

13C-PFOA Found (ng/mL) = (peak area - intercept) / slope x DF

Recovery (%) =

x 100 13C-PFOA found (ng/mL) amount 13C-PFOA added (ng/mL)

Standard Curve: Linear (1/x weighted)

Intercept = 4633.04Slope = 47928.5 Coef. Of Det. = 0.996972

CS = Calibration standard

C = Control sample

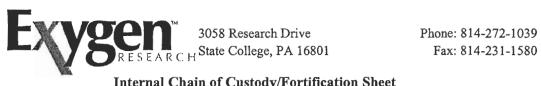
LF = Lab fortified sample LCS = Laboratory Control Spike ND = Not detected = Response between 0 and 0.05 ng/mL

S = Sample

*Sample was analyzed at several dilution levels in this set. The appropriate result is reported.

NQ = Not quantifiable = Response between 0.05 and 0.1 ng/mL

Spreadsheet prepared by: CEE /12 21 05



Internal	Chain o	f Custo	dv/Fo	rtification	Sheet
----------	---------	---------	-------	-------------	-------

Exygen Study Number:	<u>L6869</u>	Matrix: Air Tubes
Sponsor Study/Protocol No:	<u>NA</u>	
The samples listed below wer	e removed from refrigerator No	
Time	Date	Initials

CLIENT SAMPLE ID	EXYGEN ID NUMBER	FORTIFICATION (ng)
Sample Media Blank T1R0 A	L6869-1	1.0
Sample Media Blank T1R0 B	L6869-2	-
Sample Media Blank T1R0 C	L6869-3	-
Control 1 A	L6869-19	2.0
Control 1 B	L6869-20	-
Control 1 C	L6869-21	
Building 4 office T1R1 A	L6869-4	1.0
Building 4 office T1R1 B	L6869-5	-
Building 4 office T1R1 C	L6869-6	-
Building 4 mixing T1R2 A	L6869-7	1.0
Building 4 mixing T1R2 B	L6869-8	-
Building 4 mixing T1R2 C	L6869-9	-
Building 4 oven T1R3 A	L6869-10	1.0
Building 4 oven T1R3 B	L6869-11	-
Building 4 oven T1R3 C	L6869-12	-
Building 5 oven T1R4 A	L6869-13	1.0
Building 5 oven T1R4 B	L6869-14	-
Building 5 oven T1R4 C	L6869-15	-
Building 6 oven T1R5 A	L6869-16	1.0
Building 6 oven T1R5 B	L6869-17	-
Building 6 oven T1R5 C	L6869-18	_
-	-	-
-	-	-
-	-	-

	Spiking Solution Used	Volume Used for Spiking	Initials/Date
L6869-19	SS0010977 (100 ng/mL PFOA)	20 μL (10-20 μL micropipet)	BAL 12/02/05
L6869-19	SS0010991 (100 ng/mL 13CPFOA)	20 μL (10-20 μL micropipet)	BAK 12/02/05
L6869-1	SS0010991 (100 ng/mL 13CPFOA)	10 μL (10-20 μL micropipet)	BAK 12/02/05
L6869-4	SS0010991 (100 ng/mL 13CPFOA)	10 μL (10-20 μL micropipet)	BAK 12/02/05
L6869-7	SS0010991 (100 ng/mL 13CPFOA)	10 μL (10-20 μL micropipet)	BBN 12/02/05
L6869-10	SS0010991 (100 ng/mL 13CPFOA)	10 μL (10-20 μL micropipet)	BAL 12/02/05
L6869-13	SS0010991 (100 ng/mL 13CPFOA)	10 μL (10-20 μL micropipet)	BAN 12/02/65
L6869-16	SS0010991 (100 ng/mL 13CPFOA)	10 μL (10-20 μL micropipet)	BAK 12/02/05
-	-	-	-

Comments: The entire OV	S tube is used for the extraction.	
Analysis Summary:	Data Set: <u> 20205B</u>	Initials/Date: CEE /12/16/05
	Data Set:	Initials/Date:/
	Data Set:	Initials/Date:/
Set extraction/analysis dat	a verified by:	Date: /5/55/05/March 19, 2004/0

3058 Research Drive

Fax: 814-231-1580 Phone: 814-272-1039

State College, PA 16801

SAMPLE EXTRACTION AND ANALYSIS TRACKING SHEET

MATRIX: Air Tubes EXYGEN STUDY NUMBER: 16869

ANALYTES: PFOA and 13C-PFOA METHOD: ExM-079-007BS

PROTOCOL NUMBER: NA

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Lot #	OI DAYBEILIN	45.44	A 104 22 234		•	•				2001010100	DHK (2/0/10)	•	•		-		- 1	45227	1807H12	45256	-		·	,		CELLINOS
Reagents/	Materiais	Methanol	0.2 µm Filters			٠	•				Initials/Date			•			HPLC	Methanol	Ammonium Acetate	HPLC Water				-		Initials/Date
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Dilutions	(m/m/)	1	,	(1	((,	(,	((1	1	(,	1	l	1	((1				í
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Dilutions	(mL/mL)		1	١	1	1	i	(1	(0.1/1.0	,	1	0.1/10	ı	,	0.1/10	1	1	0.1/10	1	-	-	-		CEE
STEP	m																					\geq	'			12/02/05/12/16/05
STEP	7	_																				\rightarrow	,	,		BAK 12/02/05
STEP																						>	,		'	BAK 12/02/05
Exygen ID		L6869-1	1.6869-2	L6869-3	1.6869-19	L6869-20	L6869-21	L6869-4	\$-69897	9-69897	1-69897	8-6989T	6-69897	Te869-10	1,6869-11	1.6869-12	L6869-13	L6869-14	L6869-15	P 1-6889-16	L6869-17	L6869-18				
Client ID		Sammle Media Blank T1R0 A	Sample Media Blank T1R0 B	Samule Media Blank T1R0 C	Control 1 A	Control 1 B	Control 1 C	Building 4 office T1R1 A	Building 4 office T1R1 B	Building 4 office TIR1 C	Building 4 mixing T1R2 A	Building 4 mixing T1R2 B	Building 4 mixing T1R2 C	Building 4 oven T1R3 A	Building 4 oven T1R3 B	Building 4 oven T1R3 C	Building 5 oven T1R4 A	Building 5 oven T1R4 B	Building 5 oven T1R4 C	Building 6 oven T1R5 A	Building 6 oven T1R5 B	Building 6 oven T1R5 C	-			**Initials/Date

STEP 1: Disassemble the airtube into Fractions A, B and C. STEP 2: Add 1 mL of methanol and mix 3 times for ~ 30 seconds with ~ 10 minutes between mixing. STEP 3: Filter through a $0.2~\mu m$ filter into an autosampler vial. STEP 4: LC/MS/MS analysis STEP 5: LC/MS/MS reanalysis.

\$\$ Method of Analysis for the Determination of Ammonium Perfluorooctanoate (APFO) in OVS Airtubes by LC/MS/MS for SPI Processors Mass Balance Study. **Initials and date under each step indicates the personnel that performed this step.

COMMENTS:

Page Position: (1, 1)

Sample List: C:\MASSLYNX\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes.SPL Printed: Fri Dec 16 16:02:15 2005 Masslynx - Sample List

DF				_	_	<u>_</u>	_	_	_	—	_	_	<u>_</u>	_	_	_	_	_	_	10	-	-	_	-	_	100	_	-	_	100	<u>_</u>	_	_	-	_	—	100	_	-	~	.	—
Test ID	00	o c	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
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Conc (ng/mL)	0.05		0.5	1.0	5.0	10	20	1	-	}	-	2.0	2.0	2.0	0.05	0.1	-	1		!		į	ŀ	0.2	0.5	1	!	1	1	1	!	!	1		1.0	5.0	1		-	!	19	20
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Sample Type	Standard	Standard	Standard	Standard	Standard	Standard	Standard	Blank	Blank	Blank	Blank	တ္တ	တ္တ	ဗ္ဗ	Standard	Standard	Analyte	Analyte	Analyte	Analyte	Analyte	Analyte	Analyte	Standard	Standard	Analyte	Analyte	Analyte	Analyte	Analyte	Analyte	Analyte	Analyte	Blank	Standard	Standard	Analyte	Analyte	Analyte	Analyte	Standard	Standard
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	C120205-8, 0.05 ng/mL standard	L standard	L standard	L standard	L standard	. standard	 standard 		on A	on B	on C	Fraction A	Fraction B	Fraction C	nL standa	L standard								L standard	L standard										20205-4, 1.0 ng/mL standard	L standard					10 ng/mL standard	20205-1, 50 ng/mL standard
mple Description	0.05 ng/r	0.1 19/11	0.5 ng/m	1.0 ng/m	5.0 ng/m	10 ng/ml	50 ng/ml	ash	nk Fracti	nk Fracti	nk Fracti	ank Spk	ank Spk	ank Spk	0.05 ng/r	0.1 ng/m				=10				0.2 ng/m	0.5 ng/m	DF=100				DF=100				ash	1.0 ng/m	5.0 ng/mL	F=100				10 ng/ml	50 ng/mi
nple Des	0205-8,	0205-7,	0205-5	0205-4,	0205-3,	0205-2,	0205-1,	hanol Wa	69-1 Bla	69-2 Bla	69-3 Bla	69-19 BI	69-20 BI	69-21 BI	0205-8,	0205-7,	69-4	69-5	9-69	69-7, DF	2-69	8-69	6-69	0205-6,	0205-5,	69-10, D	69-10	69-11	69-12	69-13, D	69-13	69-14	69-15	hano! W	0205-4,	0205-3,	69-16, D	69-16	69-17	69-18	C120205-2, 10 no	.0205-1,
Sa	25	25	25	C12	C12	C12	C12	Met	L68	F97	R97	F 1	F97	F98	C12	C12	F97	F97	F 68	L68	F-68	F98	F 1	C12	C12	L68	F 1	F93	R97	R97	L68	F93	F98	Met	C12	C12	F 68	F93	F 1	F9	C12	ZI.S
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Je	101	102	-104	-105	106	1-107	-108	109	-110	-111	-112	1113	114	-115	-116	1117	-118	-119	-120	1-121	122	1-123	1-124	-125	1-126	1-127	1-128	129	1-130	-131	132	1-133	134	-135	-136	-137	-138	-139	-140	-141	-142	-143
File Name	120205B-101	120203B-102	120205B-104	120205B-105	120205B-	120205B-107	120205B-108	120205B-109	120205B-110	120205B-11	120205B-1	120205B-1	120205B-1	120205B-1	120205B-1	120205B-1	120205B-1	120205B-119	120205B-120	120205B-12	120205B-122	120205B-123	120205B-124	120205B-125	120205B-126	120205B-127	120205B-128	120205B-129	120205B-1	120205B-13	120205B-132	120205B-133	120205B-134	120205B-135	120205B-136	120205B-137	120205B-138	120205B-139	120205B-140	120205B-14	120205B-14	120205B-1
Vial	4 5	7 47	4	45	46	47	48	95	11	12	13	14	15	16	41	45	17	18	19	20	21	22	23	43	4	24	25	56	27	28	59	30	31	92	45	46	32	33	34	35	47	4δ
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Page 2	Page Position: (2, 1)
	Pag

Masslynx - Sample List

Sample List: C:\MASSLYNX\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes.SPL Printed: Fri Dec 16 16:02:15 2005

CEE 12/11/05

e Inj.Volume	_	_	Υ-	Ψ-	_	_	_	_	Υ-	_	Υ-	Ψ-	Ψ-	_	Ψ,	Γ,		_	_	Ψ,	_	_	~		_	_	_ ,	_ '	_ ,	_ ,		0 to	· —	_	_	~	_	_	_	_	•	0 8
MS Tune File	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorocnems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochems	Fluorochomo	
HPLC Method	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	water	10401	200
MS Method	13CP	PFOA 13CPFOA	A 13CP	S	CPF	13CPF	13CPF	13CPF	13CPF	PFOA 13CPFOA	130	PFOA 13CPFOA	PFOA 13CPFOA	PFOA 13CPFOA	PFOA 13CPFOA	PEOA 13CPFOA	PFOA 13CPFOA	PEOA 13CPEOA	PFOA 13CPFOA	PEDA 13CPEDA		PFOA 13CPFOA	130	13CP	130	OA 13CPF	13CPF	13CPF	4 13CPI	S	DEON 120 DEON											
	_	7	က	4	IJ	9	7	œ	6	10	7	12	13	14	15	16	17	9	19	20	7	22	23	24	22	56	27	8 8	62	200	بر در	33.	34	35	36	37	38	33	40	41	42	

LC/MS/MS SYSTEM AND OPERATING CONDITIONS

Sponsor Protocol No: NA

Exygen Study No: L6869

Instrument:

Micromass Quattro Ultima (LC/MS/MS Unit #6)

Computer:

COMPAQ Professional Workstation AP200

Software:

Microsoft Windows NT: Version 4 Build 1381: Service Pack 5

Micromass Limited: MassLynx 3.4 Build 004

HPLC Equipment:

Hewlett Packard (HP) Series 1100

HP Quat Pump

HP Vacuum Degasser

HP Autosampler

HP Column Oven

HPLC Column:

Genesis C-8, 5 cm x 2.1 mm i.d. x 4 μ (Exygen ID: MA0021031)

(JONESCHROMATOGRAPHY: Part No. FK5962E)

Mobile Phase (A):

2 mM Ammonium Acetate in Water

Mobile Phase (B):

Methanol

Analyst:

Chrissy Edwards CEE 12/16/05

Exygen Research

3058 Research Drive, State College, PA 16801 Phone: (814) 272-1039 FAX: (814) 231-1580

NOTE: The next 3 pages are computer generated printouts from the masslynx software program. The pages contain the instrument settings used for the analysis of this data set.

All Handwritten Peak ID's by: (FE 12 21 05

Scanning Method Report

Method: C:\MASSLYNX\FLUOROCHEMICALS.PRO\ACQUDB\PFOA 13CPFOA
Last Modified: Wed Nov 23 14:39:10 2005

Printed: Fri Dec 16 16:02:32 2005

Solvent Delay (mins): 0.00

Analog Channel 4: Unused Function: 1 MRM of 2 Mass Pairs (ESP-)

Inter Channel Delay (Secs) : 0.03
Span (Daltons) : 0.00
Start Time (Mins) : 0.00
End Time (Mins) : 8.00
Repeats : 1

Channel Parent Daughter Dwell (Secs) Coll Energy (eV) Cone (V)

1 413.00 369.00 0.20 10 10 2 415.00 370.00 0.20 10 10 Method Report

Method File: Last Modified: c:\masslynx\fluorochemicals.pro\acqudb\water
Friday, December 16, 2005 15:46:34

Printed:

Friday, December 16, 2005 16:02:44

CEE 12/16/05

HP1100 LC Pump Initial Conditions

Solvents	
A%	60.0
B%	40.0
C%	0.0
D%	0.0
Flow (ml/min)	0.300
Stop Time (mins)	15.0
Min Pressure (bar)	0
Max Pressure (bar)	400
Oven Temperature Left(°C)	35.0
Oven Temperature Right(°C)	35.0

HP1100 LC Pump Gradient Timetable

The gradient Timetable contains 10 entries which are :

Time	A%	B%	С¥	D%	Flow	Pressure
0.00	60.0	40.0	0.0	0.0	0.300	400
0.40	60.0	40.0	0.0	0.0	0.300	400
1.00	10.0	90.0	0.0	0.0	0.300	400
7.00	10.0	90.0	0.0	0.0	0.300	400
7.50	0.0	100.0	0.0	0.0	0.300	400
9.00	0.0	100.0	0.0	0.0	0.400	400
9.50	60.0	40.0	0.0	0.0	0.400	400
13.50	60.0	40.0	0.0	0.0	0.400	400
14.00	60.0	40.0	0.0	0.0	0.300	400
15.00	60.0	40.0	0.0	0.0	0.300	400

HP1100 LC Pump External Event Timetable

The Timetable contains 6 entries which are :

Time	Column Switch	Contact1	Contact2	Contact3	Contact4
Initial	Off	Off	Off	Off	Off
0.00	Off	On	Off	Off	Off
0.05	Off	Off	Off	Off	Off
0.10	Off	Off	On	Off	Off
7.95	Off	Off	Off	On	Off
8.00	Off	Off	Off	Off	Off

HP1100 Autosampler Initial Conditions

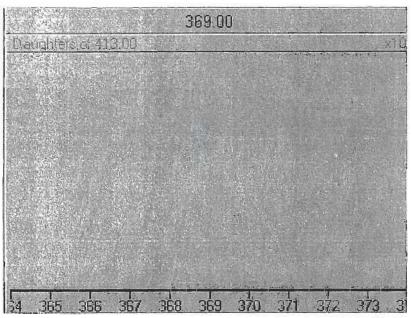
Draw Speed	200.0
Eject Speed (µl/min)	200
Draw Position (mm)	0.50
Stop Time (mins)	15.00
Injection Volume (μl)	15.0
Vial Number	93

Tuning Method Report

Page 1

C:\MASSLYNX\FLUOROCHEMICALS.PRO\ACQUDB\FLUOROCHEMS

Printed: Fri Dec 16 16:02:57 2005



Dau 413.00

SOURCE (ESP-)	Set	Rdbk	Analyser	Set	Rdbk
Capillary Cone Hexapole 1 Aperture 1 Hexapole 2 Source Block Temp. Desolvation Temp.	3.00 10 0.0 0.0 0.0 100 300	-2.93 -10	LM Res 1 HM Res 1 IEnergy 1 Entrance Collision Exit LM Res 2 HM Res 2 IEnergy 2 Multiplier	13.5 13.5 1.0 -2 30 2 13.5 13.5 2.0 650	8 9 11
Pressures	Rdbk		Gas Flows	Rdbk	
Analyser Vacuum Gas Cell	OFF 3.0e-	3	Cone Gas Desolvation	188.6 726.4	

Quantify Calibration Report

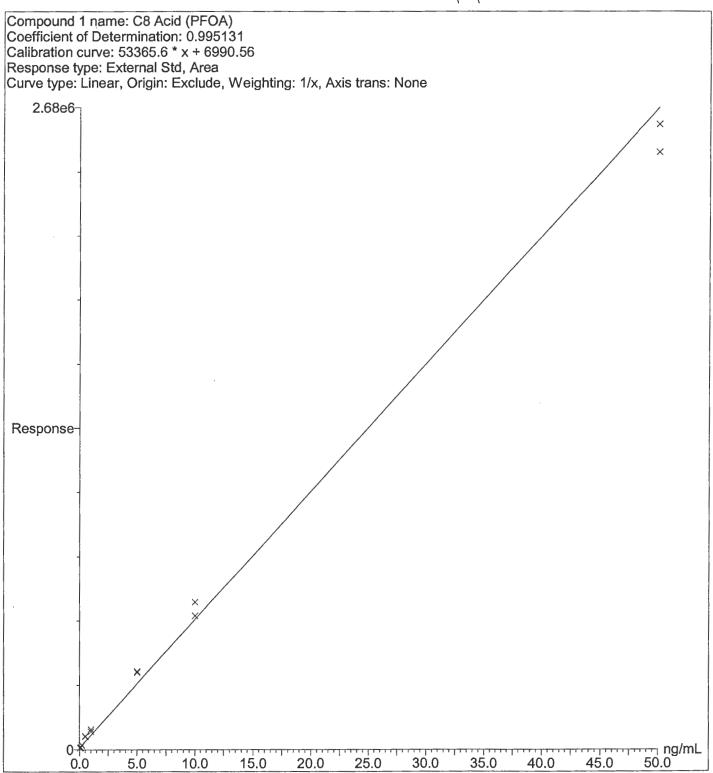
Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Page 1

P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\CurveDB\120205B OVS Tubes

Last modified: Wed Dec 21 12:45:40 2005

Printed: Wed Dec 21 12:46:00 2005

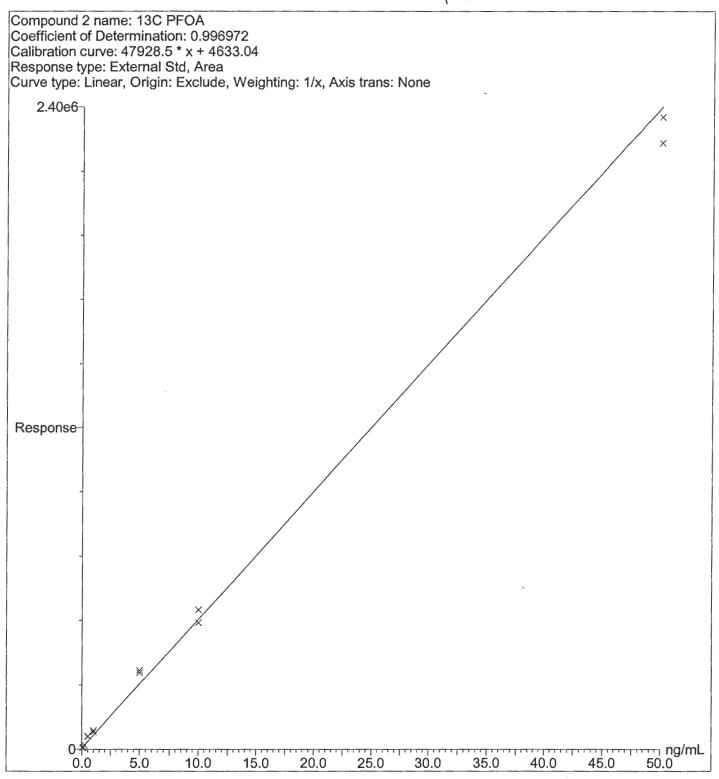


Quantify Calibration Report Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards Page 2

P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\CurveDB\120205B OVS Tubes Calibration:

Last modified: Wed Dec 21 12:45:40 2005

Printed: Wed Dec 21 12:46:00 2005



Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Page 1

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

Job Code:

Printed:

Wed Dec 21 12:46:01 2005

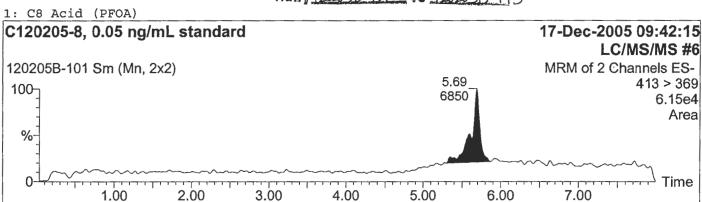
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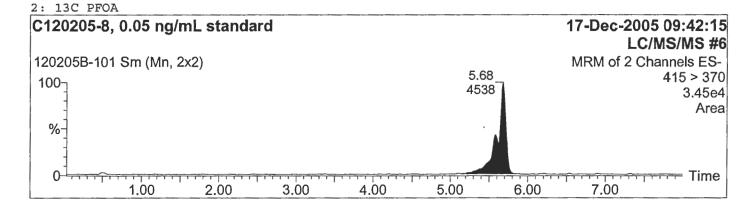
Text:

Initials CEE

Date 12|21|05

Run#120205B-101 **To** 120205B-143





Page 2

Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

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Last modified: Tue Nov 29 08:48:19 2005

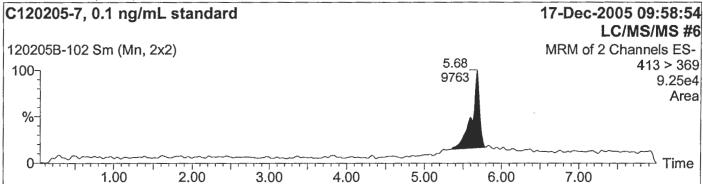
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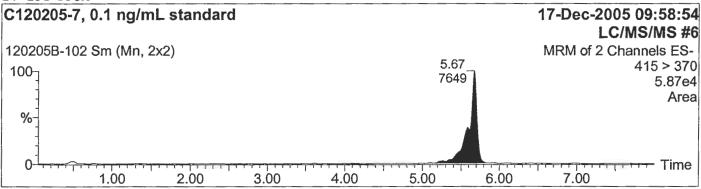
Wed Dec 21 12:46:01 2005

Name: 120205B-102

Text:







Study No.: L6869, Set No.: 120205B, Ext.Date: 12/02/05, Analyst: C.Edwards

Page 3

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Last modified: Mon Dec 19 15:41:41 2005

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Last modified: Tue Nov 29 08:48:19 2005

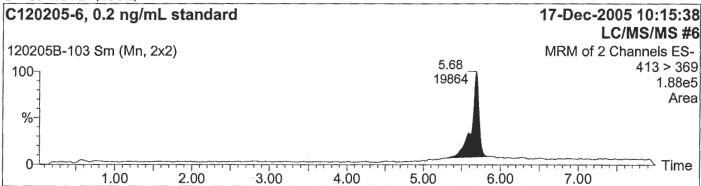
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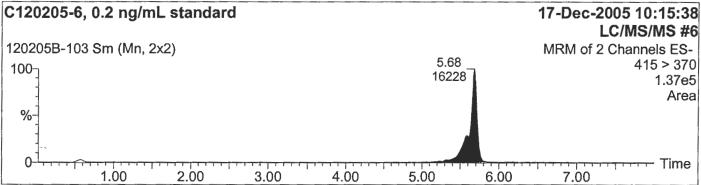
Wed Dec 21 12:46:01 2005

Name: 120205B-103

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Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Page 4

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

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Last modified: Tue Nov 29 08:48:19 2005

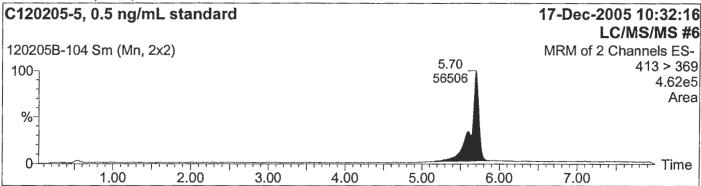
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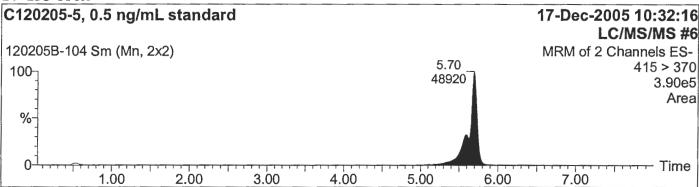
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Name: 120205B-104

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Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Page 5

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

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Last modified: Tue Nov 29 08:48:19 2005

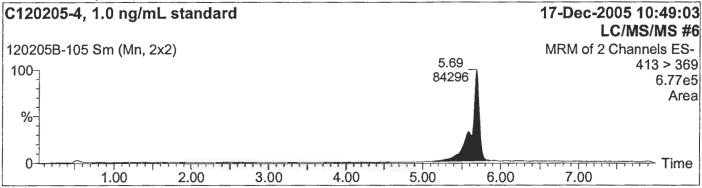
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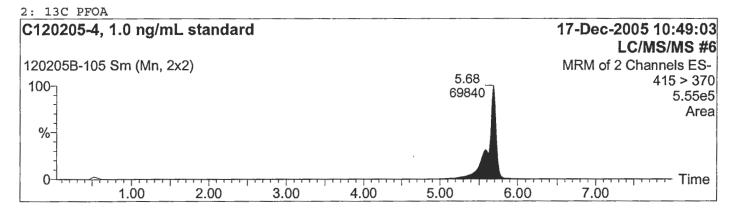
Printed:

Wed Dec 21 12:46:01 2005

Name: 120205B-105

Text:





Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Page 6

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

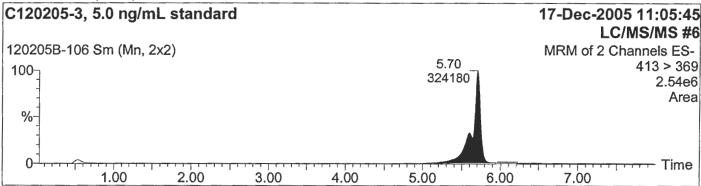
Job Code:

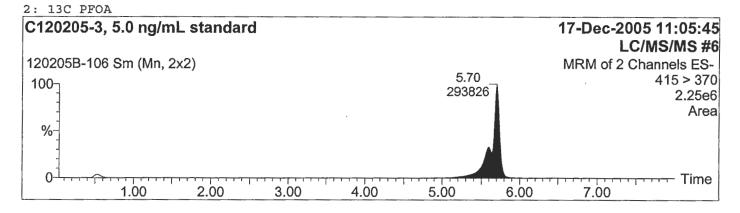
Printed:

Wed Dec 21 12:46:01 2005

Name: 120205B-106

Text:





Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

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Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

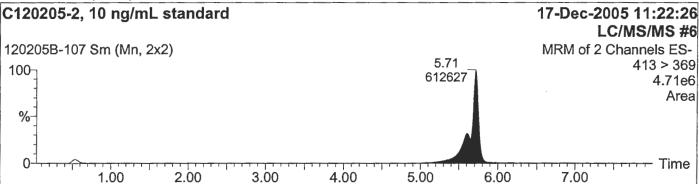
Job Code:

Printed:

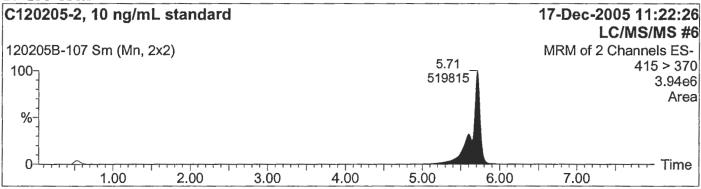
Wed Dec 21 12:46:01 2005

Name: 120205B-107

Text:







Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

Job Code:

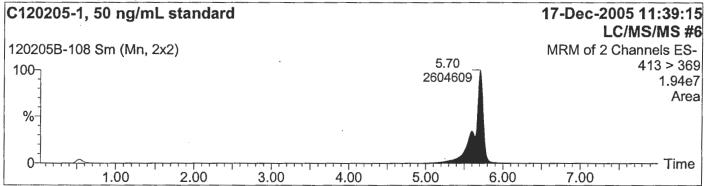
Printed:

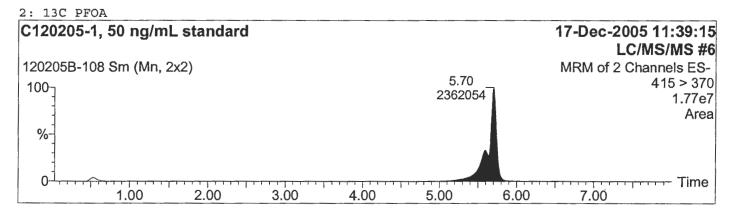
Wed Dec 21 12:46:01 2005

Name: 120205B-108

Text:

1: C8 Acid (PFOA)





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Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

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Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

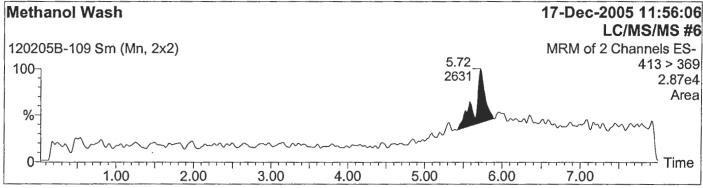
Job Code:

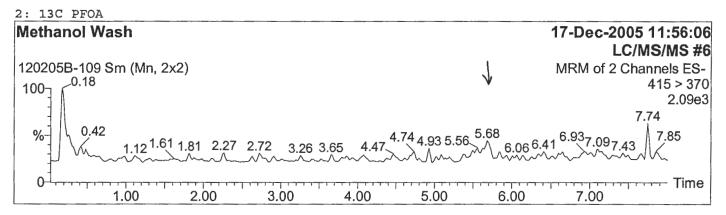
Printed:

Wed Dec 21 12:46:01 2005

Name: 120205B-109

Text:





Quantify Sample Report Page 10

Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

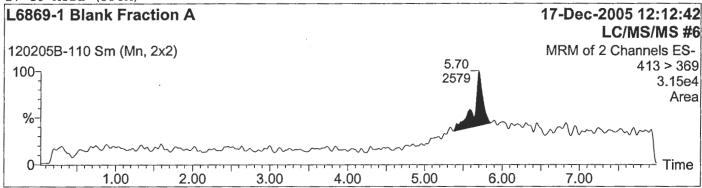
Job Code:

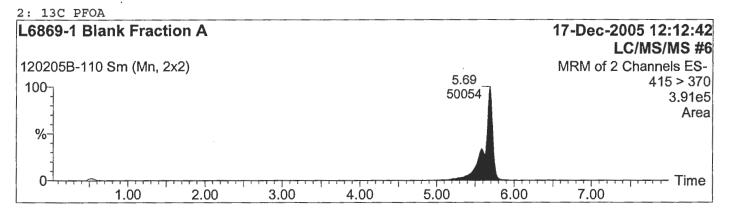
Printed:

Wed Dec 21 12:46:01 2005

Name: 120205B-110

Text:





Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

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Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

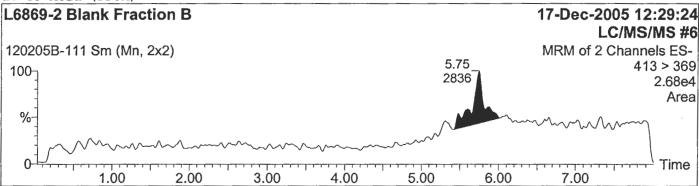
Job Code:

Printed:

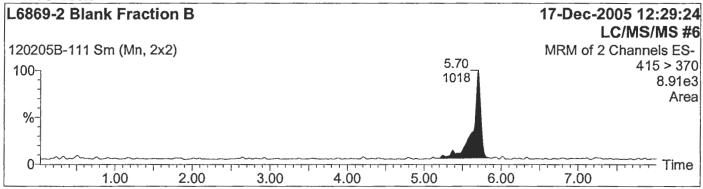
Wed Dec 21 12:46:01 2005

Name: 120205B-111

Text:







Quantify Sample Report Page 12

Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

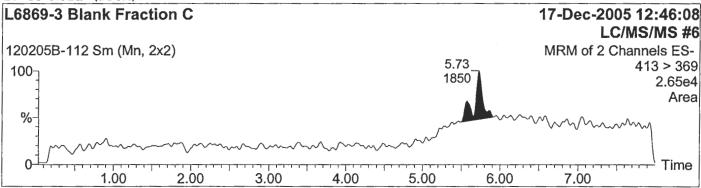
Job Code:

Printed:

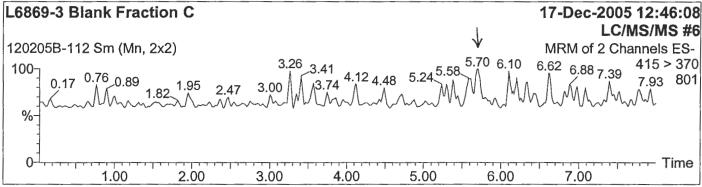
Wed Dec 21 12:46:01 2005

Name: 120205B-112

Text:







Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

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Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

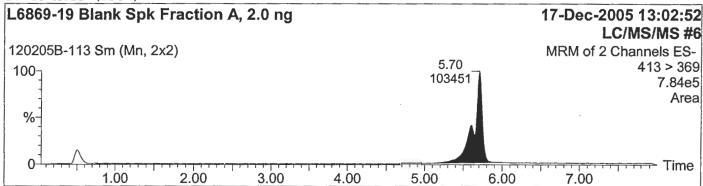
Job Code:

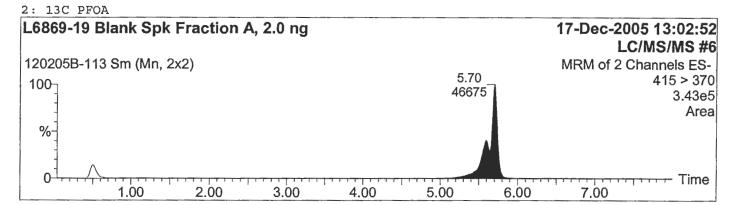
Printed:

Wed Dec 21 12:46:01 2005

Name: 120205B-113

Text:





Quantify Sample Report Page 14

Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

Job Code:

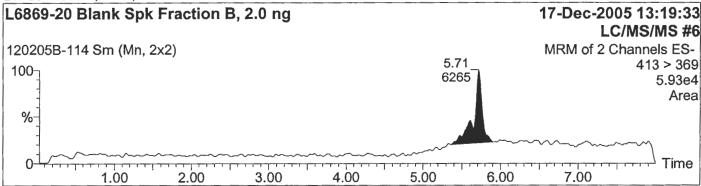
Printed:

Wed Dec 21 12:46:01 2005

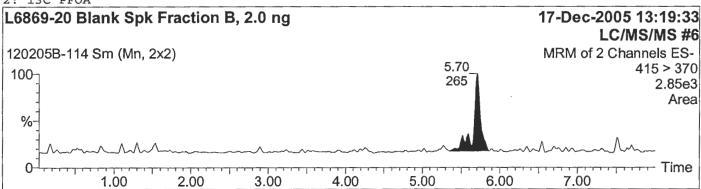
Name: 120205B-114

Text:

1: C8 Acid (PFOA)



2: 13C PFOA



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Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

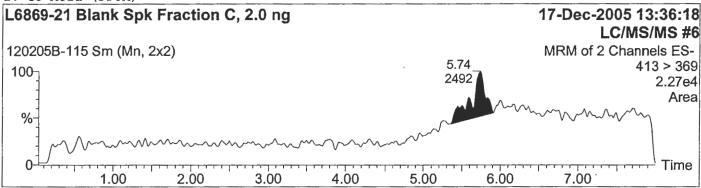
Job Code:

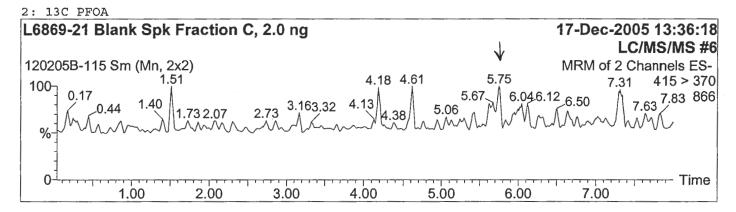
Printed:

Wed Dec 21 12:46:01 2005

Name: 120205B-115

Text:





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Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

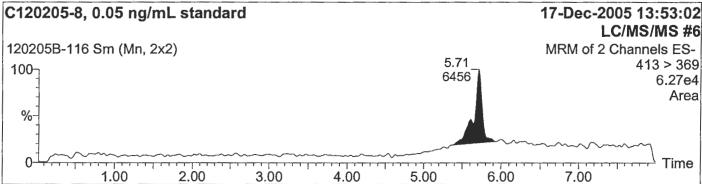
Job Code:

Printed:

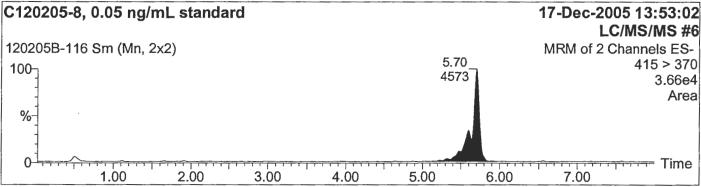
Wed Dec 21 12:46:01 2005

Name: 120205B-116

Text:







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Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1

Last modified: Tue Nov 29 08:48:19 2005

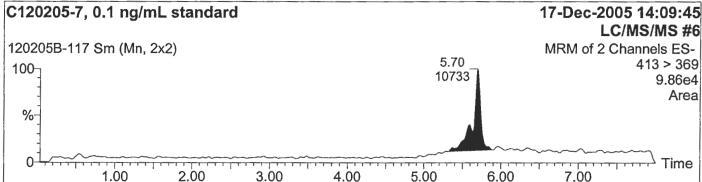
Job Code:

Printed:

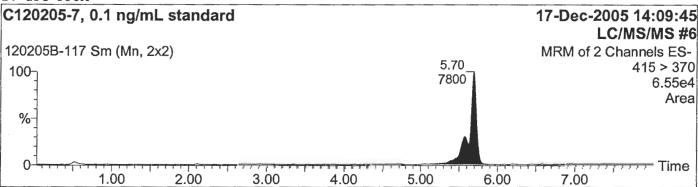
Wed Dec 21 12:46:01 2005

Name: 120205B-117

Text:







Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

Job Code:

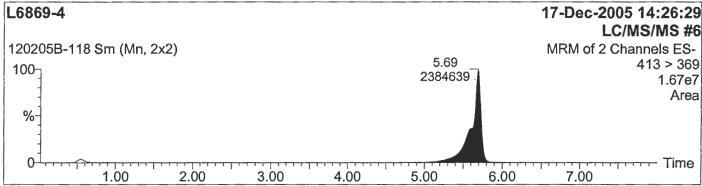
Printed:

Wed Dec 21 12:46:01 2005

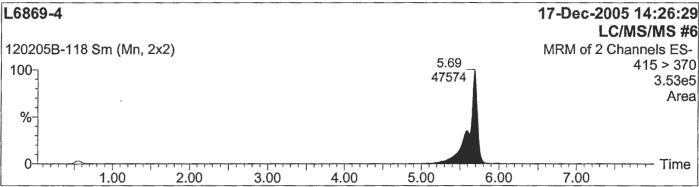
Name: 120205B-118

Text:

1: C8 Acid (PFOA)







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Quantify Sample Report Page 19

Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

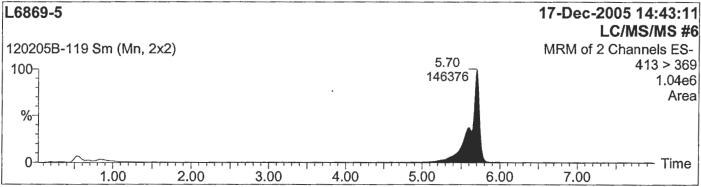
Job Code:

Printed:

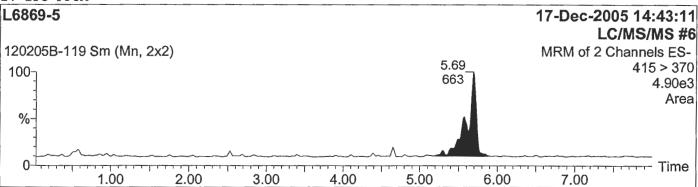
Wed Dec 21 12:46:01 2005

Name: 120205B-119

Text:







Quantify Sample Report Page 20

Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

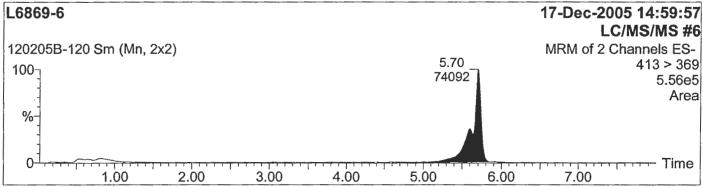
Job Code:

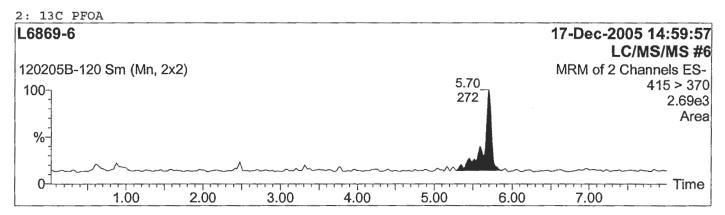
Printed:

Wed Dec 21 12:46:01 2005

Name: 120205B-120

Text:





Study No.: L6869, Set No.: 120205B, Ext.Date: 12/02/05, Analyst: C.Edwards

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Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1

Last modified: Tue Nov 29 08:48:19 2005

Job Code:

Printed:

Wed Dec 21 12:46:01 2005

Name: 120205B-121

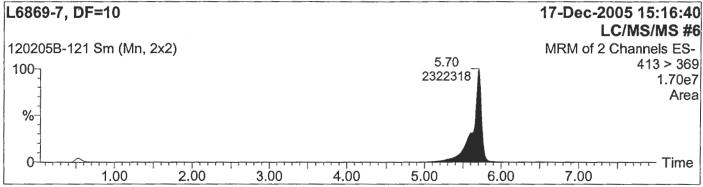
Text:

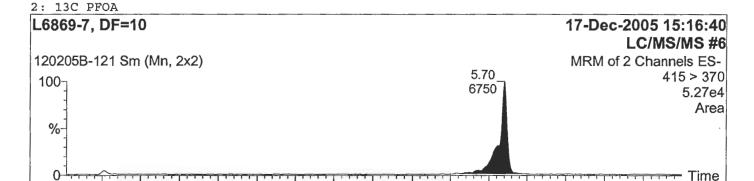
1: C8 Acid (PFOA)

1.00

2.00

3.00





4.00

5.00

6.00

7.00

Study No.: L6869, Set No.: 120205B, Ext.Date: 12/02/05, Analyst: C.Edwards

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Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

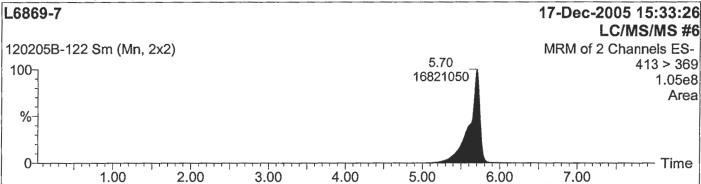
Job Code:

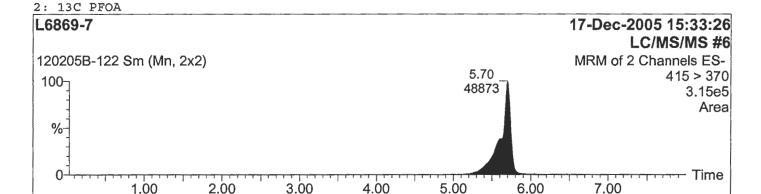
Printed:

Wed Dec 21 12:46:01 2005

Name: 120205B-122

Text:





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Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

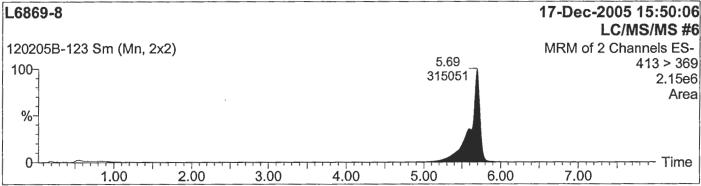
Job Code:

Printed:

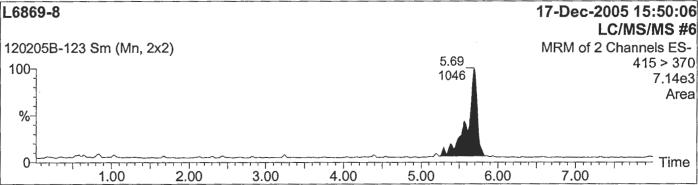
Wed Dec 21 12:46:01 2005

Name: 120205B-123

Text:







Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

Job Code:

Printed:

Wed Dec 21 12:46:01 2005

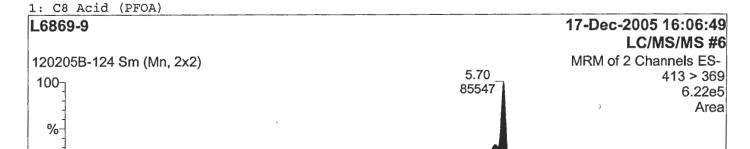
Name: 120205B-124

1.00

2.00

3.00

Text:

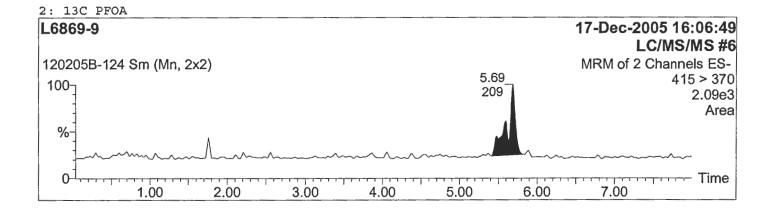


4.00

5.00

6.00

7.00



Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

Job Code:

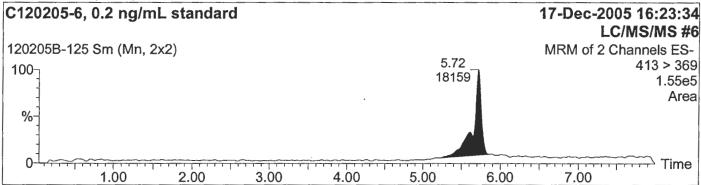
Printed:

Wed Dec 21 12:46:01 2005

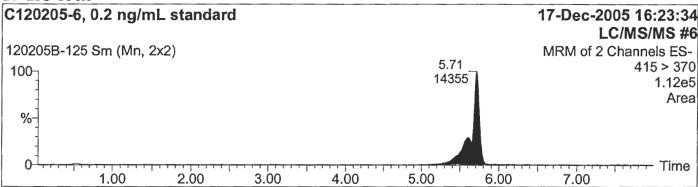
Name: 120205B-125

Text:

1: C8 Acid (PFOA)







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Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

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Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

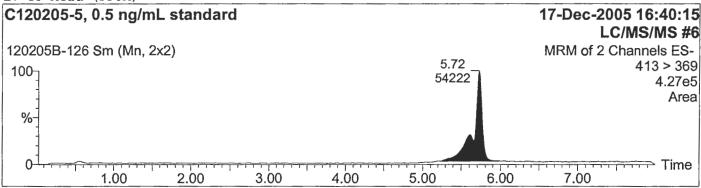
Job Code:

Printed:

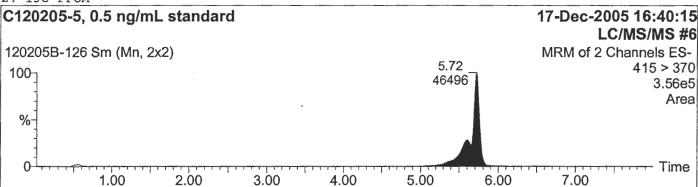
Wed Dec 21 12:46:01 2005

Name: 120205B-126

Text:







Quantify Sample Report Page 27

Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

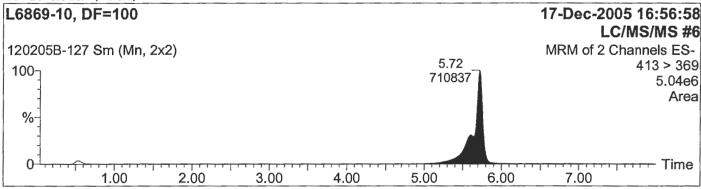
Job Code:

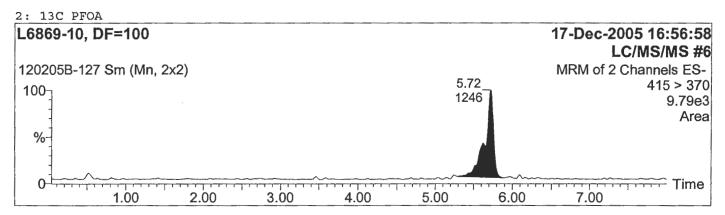
Printed:

Wed Dec 21 12:46:01 2005

Name: 120205B-127

Text:





Study No.: L6869, Set No.: 120205B, Ext.Date: 12/02/05, Analyst: C.Edwards

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Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

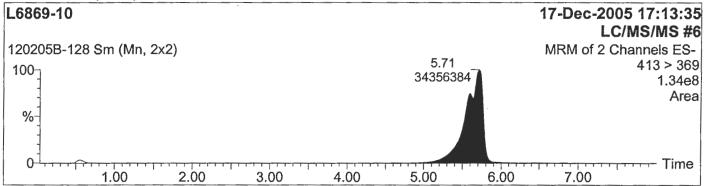
Job Code:

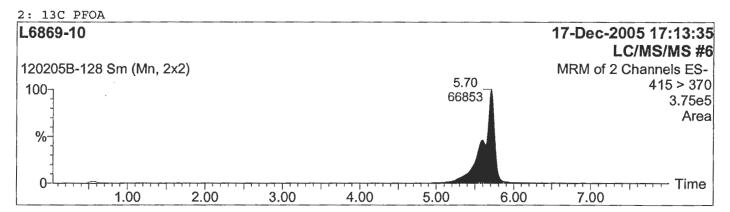
Printed:

Wed Dec 21 12:46:01 2005

Name: 120205B-128

Text:





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Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

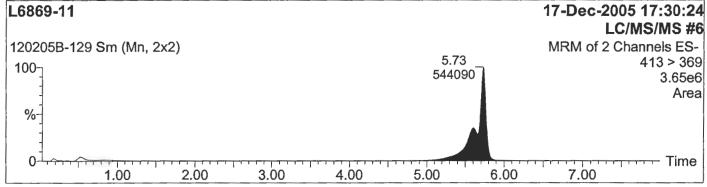
Job Code:

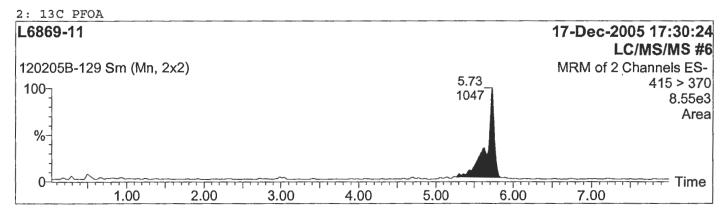
Printed:

Wed Dec 21 12:46:01 2005

Name: 120205B-129

Text:





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Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

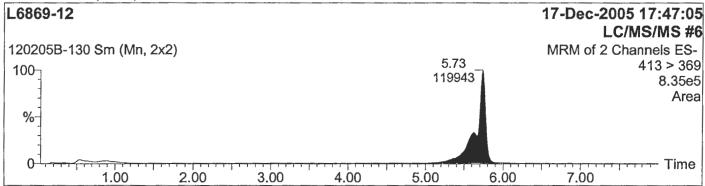
Job Code:

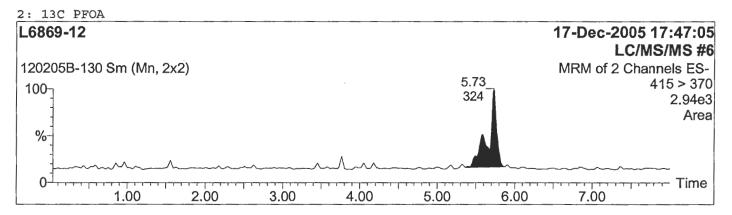
Printed:

Wed Dec 21 12:46:01 2005

Name: 120205B-130

Text:





Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Page 31

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

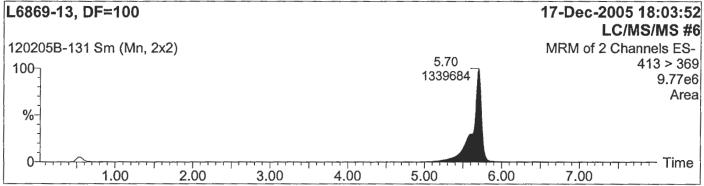
Job Code:

Printed:

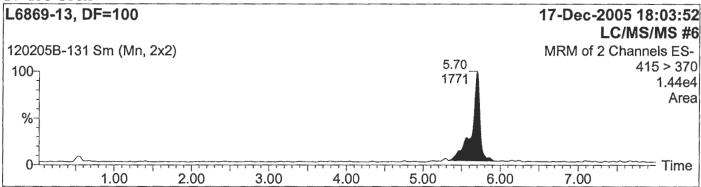
Wed Dec 21 12:46:01 2005

Name: 120205B-131

Text:







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Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

Job Code:

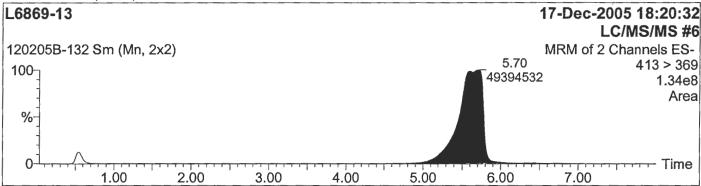
Printed:

Wed Dec 21 12:46:01 2005

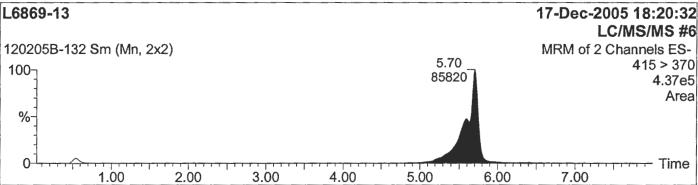
Name: 120205B-132

Text:









Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Page 33

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1

Last modified: Tue Nov 29 08:48:19 2005

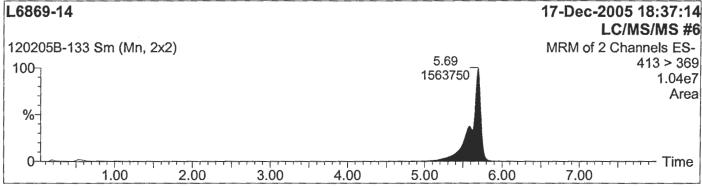
Job Code:

Printed:

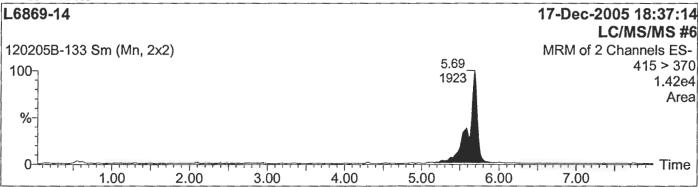
Wed Dec 21 12:46:01 2005

Name: 120205B-133

Text:







Study No.: L6869, Set No.: 120205B, Ext. Date: 12/02/05, Analyst: C. Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

Job Code:

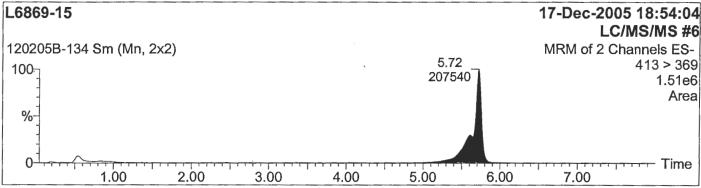
Printed:

Wed Dec 21 12:46:01 2005

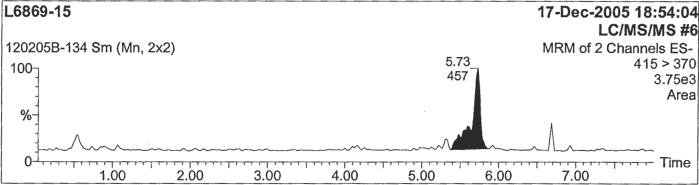
Name: 120205B-134

Text:

1: C8 Acid (PFOA)







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Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Page 35

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

Job Code:

Printed:

Wed Dec 21 12:46:01 2005

Name: 120205B-135

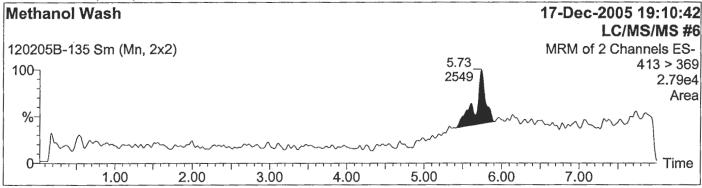
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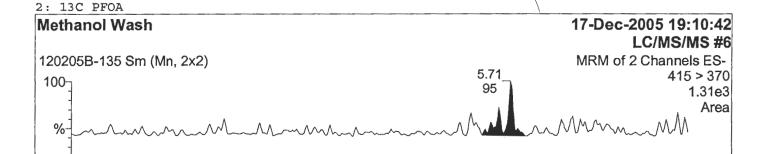


1.00

2.00

3.00





4.00

5.00

→ Time

6.00

7.00

Study No.: L6869, Set No.: 120205B, Ext. Date: 12/02/05, Analyst: C. Edwards

Page 36

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

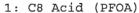
Job Code:

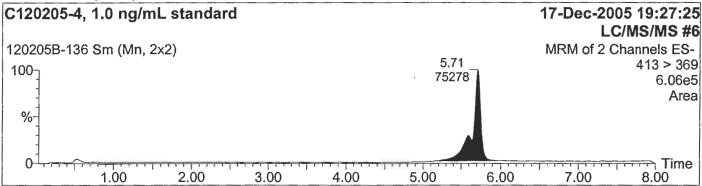
Printed:

Wed Dec 21 12:46:01 2005

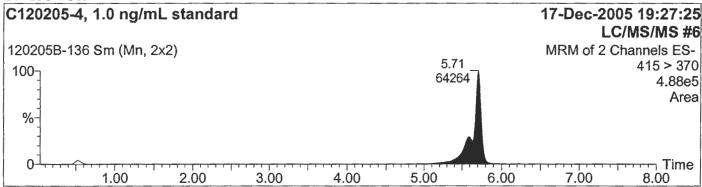
Name: 120205B-136

Text:





2: 13C PFOA



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Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

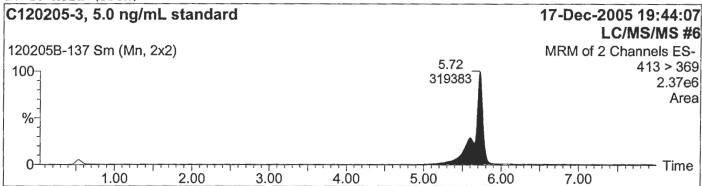
Job Code:

Printed:

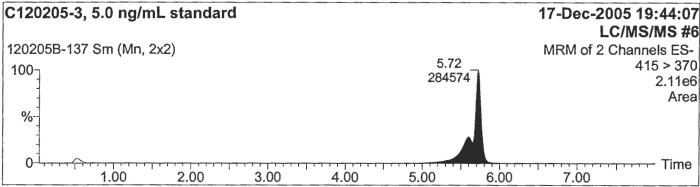
Wed Dec 21 12:46:01 2005

Name: 120205B-137

Text:







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Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

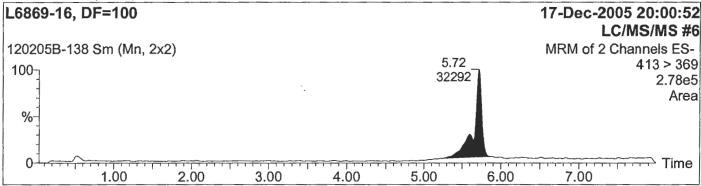
Job Code:

Printed:

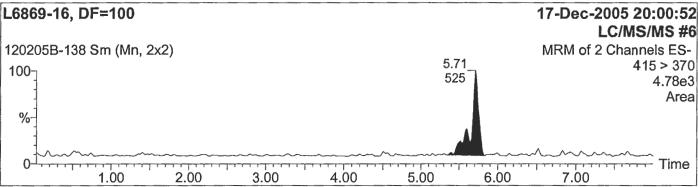
Wed Dec 21 12:46:01 2005

Name: 120205B-138

Text:







Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

Job Code:

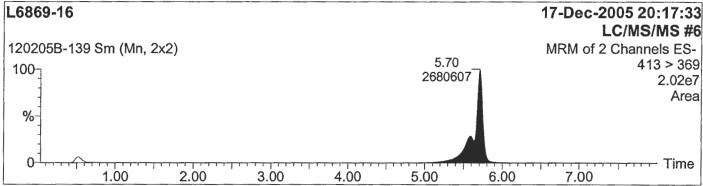
Printed:

Wed Dec 21 12:46:01 2005

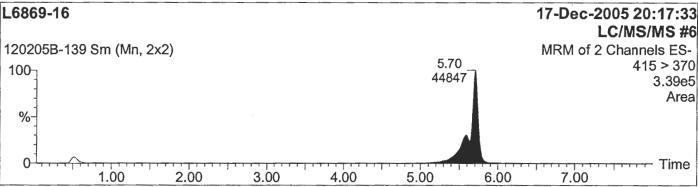
Name: 120205B-139

Text:

1: C8 Acid (PFOA)







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Study No.: L6869, Set No.: 120205B, Ext.Date: 12/02/05, Analyst: C.Edwards

P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

Job Code:

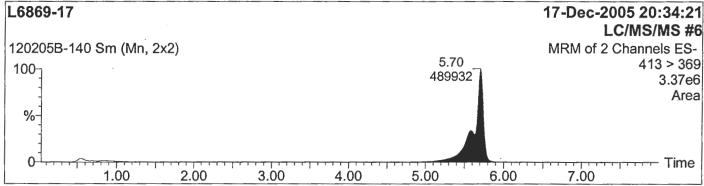
Printed:

Wed Dec 21 12:46:01 2005

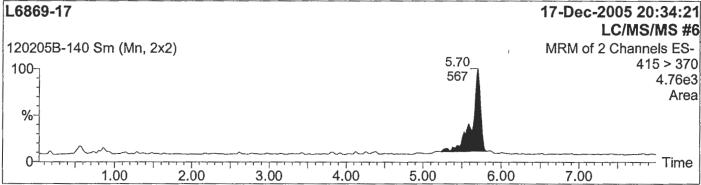
Name: 120205B-140

Text:

1: C8 Acid (PFOA)







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Page 41

Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

Job Code:

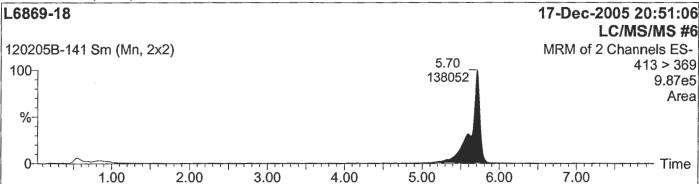
Printed:

Wed Dec 21 12:46:01 2005

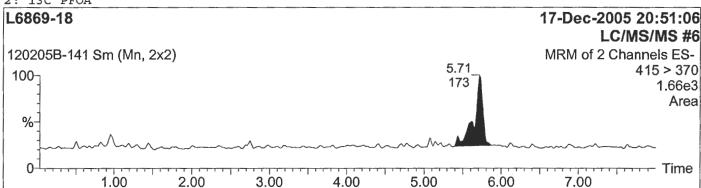
Name: 120205B-141

Text:





2: 13C PFOA



Page 42 Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

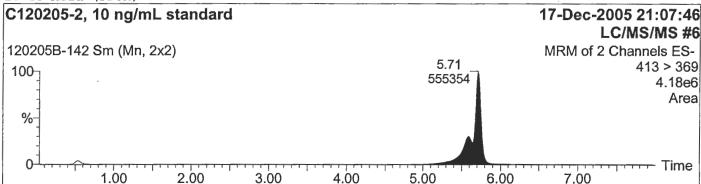
Job Code:

Printed:

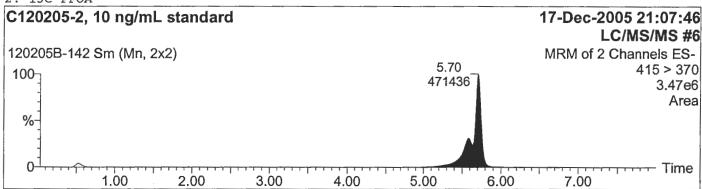
Wed Dec 21 12:46:01 2005

Name: 120205B-142

Text:







Study No.:L6869, Set No.:120205B, Ext.Date:12/02/05, Analyst: C.Edwards

Page 43

Sample List: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\SampleDB\120205B OVS Tubes

Last modified: Mon Dec 19 15:41:41 2005

Method: P:\Data\LCMSMS6\Masslynx\Fluorochemicals.PRO\MethDB\PFOA 13C PFOA ng-mL 1:

Last modified: Tue Nov 29 08:48:19 2005

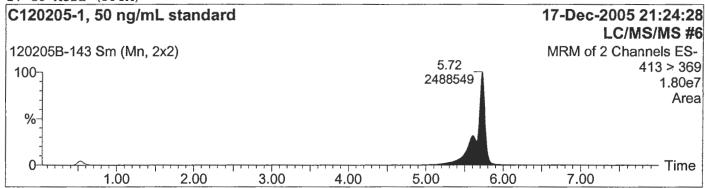
Job Code:

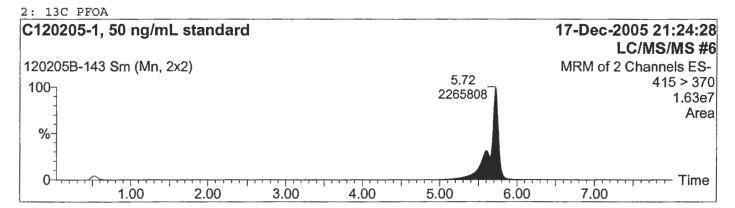
Printed:

Wed Dec 21 12:46:01 2005

Name: 120205B-143

Text:





Appendix C

Field Data Sheets



HiFlow Pumps

Project Name Project No.	32/01-002-00			Building 4 lunch room	
THE RESERVE THE PERSON NAMED IN COLUMN TWO IS NOT THE PERSON NAMED IN COLUMN TWO IS NAMED IN COLUMN TWO I	SKC 724-PC		Serial Number		
BIOS DryCal	DC-2 Flow Calib	orator	Serial Number	B 924	
	Calibration Rate liters/min	Pump Rotometer liters/min	Tech / Notes		
Pretest	1.008	1.0	Performed on 1/21/05 - enected 11/12 AM		
Posttest	0.9993	1.0	Performed on	Mazlos PM	
Sample Media	a	Sample ID			
Leak	Start	Stop	Elapsed Time	Tech	
Check	Time	Time	(minutes)	Initials	
O.K.	0853	1553	420	TMIZ	
Time	Observations				
0853	Sampler located above Refrigerator in Lunchroom				
1015				resent. No safety glasses	
11/8	145 min/Flow O.K. No employees present.				
1212	Flowork 4 employees Present.				
1331	278 min/ Zemployees Present.				
1418	Flowax Sovemployees present.				
1515	382min/ No employees				
1553	womin /			ec) Present	
		A CARLO A STATE OF THE STATE OF			
				2.11	
COLUMN TO THE PARTY OF THE PART		CONTRACTOR OF A CONTRACTOR OF	THE REST CO. THE REAL PROPERTY AND ADDRESS OF THE PERSON.	A CONTRACTOR OF THE PARTY OF TH	
		CALL SERVICE STATE SANDERS AND SERVICE STATE OF SERVICE SANDERS AND SERVICE SANDERS AN	TATALON OF STREET		
opcount minimum, accommode must be		The street of th		and the second s	
Sub-international contractions of the contraction o		ATTENDED TO SECURITION OF THE	The state of the s	THE COLUMN TWO IS NOT THE OWNER. THE COLUMN TWO IS NOT THE OWNER.	
			AND DESCRIPTION OF THE PERSON	The second section of the second	



HiFlow Pumps

Project Name	whiteman os	kman at Hanna	ພະ Test Date	11-22-05 TIRZ	
	32/01-002-00			Mixing Room- Duilding 4	
	SKC 224-PC		Serial Number	707998	
BIOS DryCal I	OC-2 Flow Calib	orator	Serial Numbe		
	Calibration Rate liters/min	Pump Rotometer liters/min	Cocated in room nearest over operations, Tech / Notes		
Pretest	1.011	1.0	performed on 11/21/05 - enerted Va An		
Posttest	0.9837	1.0	performed on 11/22/05 PM		
Sample Media	1	Sample ID			
Leak	Start	Stop	Elapsed Time	Tech	
Check	Time	Time	(minutes)	Initials	
O.K.	0857	1558	421	TMR	
Time	Observations				
0857	Sampler 1	ocated on	Table/workst	gtion serving room	
ipen MB		MANAMO	nsymv	MA FIGURE	
1022	BYMIN JUNE / No employees Present. Two mixing stations operating				
1120	Flower / 1 employee Present. Three mixing stayions operating				
1214	Flow o.K. / I'cm Player Present / Two mixing c totions oftent in				
1733	Flow O.K. No employees / Three mixing stations				
1422	Flowork.	No emplo	1ees / T	hree mixing stations	
1517	Blowok /	No Emplo	yees In	ree mixing stations	
1558	421 Min (1 employ		re mixing stations	
			and the second		



HiFlow Pumps

Project Name	Whiteman, Oste	vaun of Hinna L	€ Test Date	11-22-05 TR3	
Project No.					
Pump Model					
BIOS DryCal DC-2 Flow Calibrator Serial Number 8924				r B924	
	Calibration Rate liters/min	Pump Rotometer liters/min	Tech / Notes		
Pretest	1.008	1.0	Parformed on 1/21/05 - checked 1/22 AM		
Posttest	0.9904	1.0		11/22/05 DM	
Sample Media	·	Sample ID			
Leak	Start	Stop	Elapsed Time	Tech	
Check	Time	Time	(minutes)	Initials	
O.K.	0902	1602	420	TMR	
Time	Observations				
0902	Located, one end of over & ~ 3' Above Dir Pan				
1018	76min Flow O.K. Ovens 8, 10, 7, 11 in operation, 5th employeesin				
112)	Flowork over 8,7,11 in o Peration, 4 employees in AreA.				
1216	Flasok O	RN 8, 11	in operatio	M. Hemployees in Area.	
1334	ELWORD oven 8,10, il in operation 4 employees				
1424	Flower over 8, 10, 18 in operation 4 employees				
1518	Fusik/ oven 8,10,11 in operation 5 employees				
1602	40 min over 8, 10, 11 in operation 3 employees				
	Sample may have been "salted" by employed carly in sample period, white flooropolymer dust sprinkled Near/Around pump site of unknown origin,				

13763



HiFlow Pumps

Project Name	Whiteman ost			11-22-05 TIR4	
Project No.	32/01-002-0	101-002 S	ample Location	Building 5 over Room	
Pump Model	SKC 224-1	PLXIZE Pump	Serial Number	761017	
BIOS DryCal I	DC-2 Flow Calib	orator	Serial Numbe	r 8 924	
	Calibration Rate liters/min	Pump Rotometer liters/min	Tech / Notes		
Pretest	1.009	1.0	Performed on "/22/05 - cheeked "YZZAM"		
Posttest	0.9869	1.0	Performed or	Yzz/os PM	
Sample Media		Sample ID			
Leak	Start	Stop	Elapsed Time	Tech	
Check	Time	Time	(minutes)	Initials	
O.K	0910	1611	421	TMR	
Time	Observations				
0910	Located between oven CF a CK				
1025	FLOW O.K/D	vens Etc, C	I, operating. C	5, ck B Being setup,	
1126	Flow O.K. OVE	as ca, ca co	I operating , ex	Sidled. & enployees Arca.	
1719	Flow o.K. overs CB, CL, CI, PK operating 4 employees in Are				
1337	Elowark/ovens CB, U CK, J. OS oferating. 3 employees.				
1426	Flow OK OVE	15 (B, CK	, cs ope	erating 3 employees	
1520			, cs ope		
1611	Gentlayers (B, CL, CK, CJ operating 6 employers				
	22-22-2-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1		, , , , , , , , , , , , , , , , , , ,		
				A STATE OF THE STA	
		PROPERTY AND ADDRESS OF THE PERSON ADDRESS OF THE PERSON AND ADDRESS OF THE PERSON AND ADDRESS OF THE PERSON AND ADDRESS OF THE PERSON ADDRESS O			



HiFlow Pumps

Project Name	Whiteman Oster	mana Uanna	ادر Test Date	11-77-05 TIRS		
A CONTRACTOR OF THE PROPERTY O	Project Name Uniteman Ostermana Hanna LLC Test Date 11-77-05 TIRS Project No. 32/01-002-001-002 Sample Location Building 6 Oven Room					
	SEC 724-174	で名 Pump	Serial Number	761019		
BIOS DryCal I	BIOS DryCal DC-2 Flow Calibrator			Serial Number B924		
	Calibration Rate liters/min	Pump Rotometer liters/min	Tech / Notes			
Pretest	1.008	1.0	performed on 1/21/05 - checked Wer AN			
Posttest	0.9967	1.0		on Wzz/05 PM		
Sample Media	1	Sample ID				
Leak	Start	Stop	Elapsed Time	Tech		
Check	Time	Time	(minutes)	Initials		
0.K	0924	1624	420	TMR		
Time	Observations					
0974	Pump Placed on over B workstand					
1043	1 13 44 64		Hing. Z Emplo			
1140	135 min / 6		,	em Ployees in Area		
1229				z employees in AreA		
1346	ZGZMA OVERS 4, B, C Operaling z employees in Area					
1487	313 min / overs A.B operating 2 employees					
1530	Howard great Bic operating Zemployers					
1624	420min/ove	ns BIC	operating !	Z imployees		
			•			
				The state of the s		
		A STATE OF THE STA		And the second of the second o		

Appendix D

Project Participants

PRIVILEGED AND CONFIDENTIAL PREPARED AT THE REQUEST OF LEGAL COUNSEL

Project Participants

Exygen Research

Jay Bernarding - Business Development Manager

Barr Engineering Company

Tim Russell - Vice President/Chemical Engineer

Beth Havlik - Vice President/Chemical Engineer

Karen Stoller - CIH

Richard Berg - Senior Air Quality Technician